



OVERVIEW

Nanopores or Ion Channels and its Applications

Types of Nanopores or Ion Channels

Nanopores or Ion Channel via Macrocycles

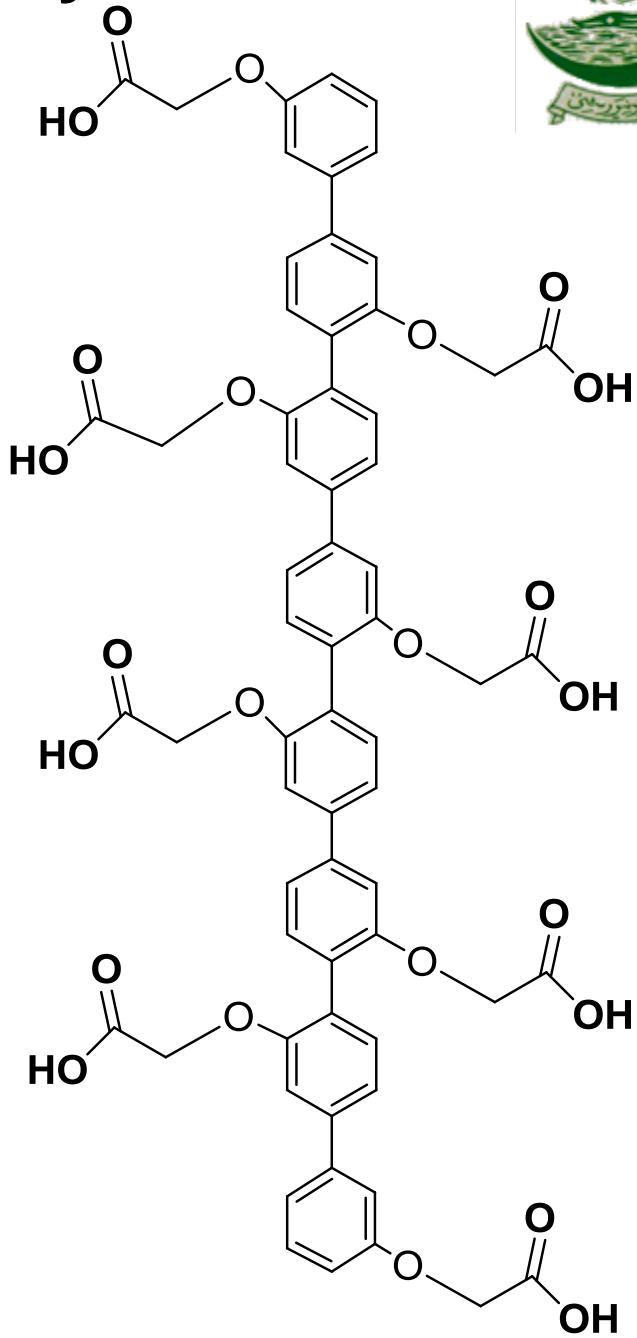
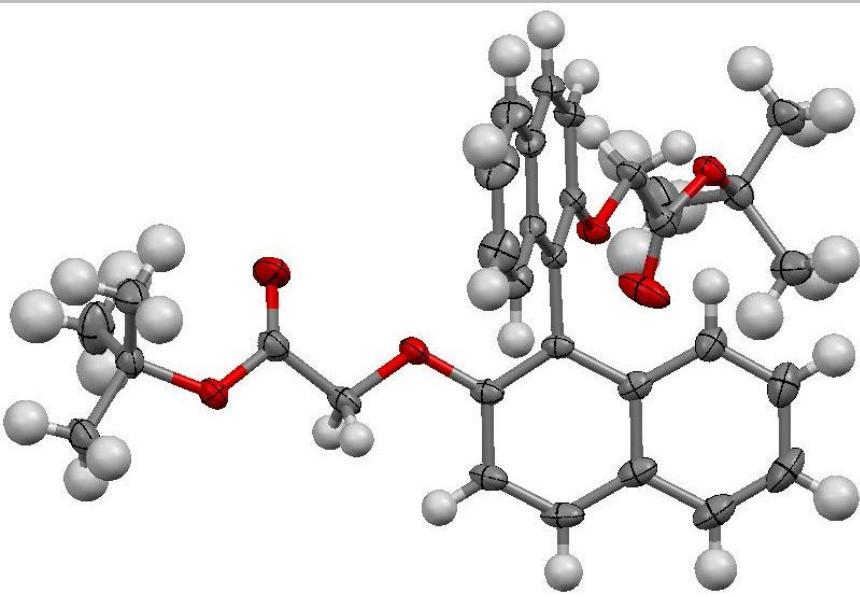
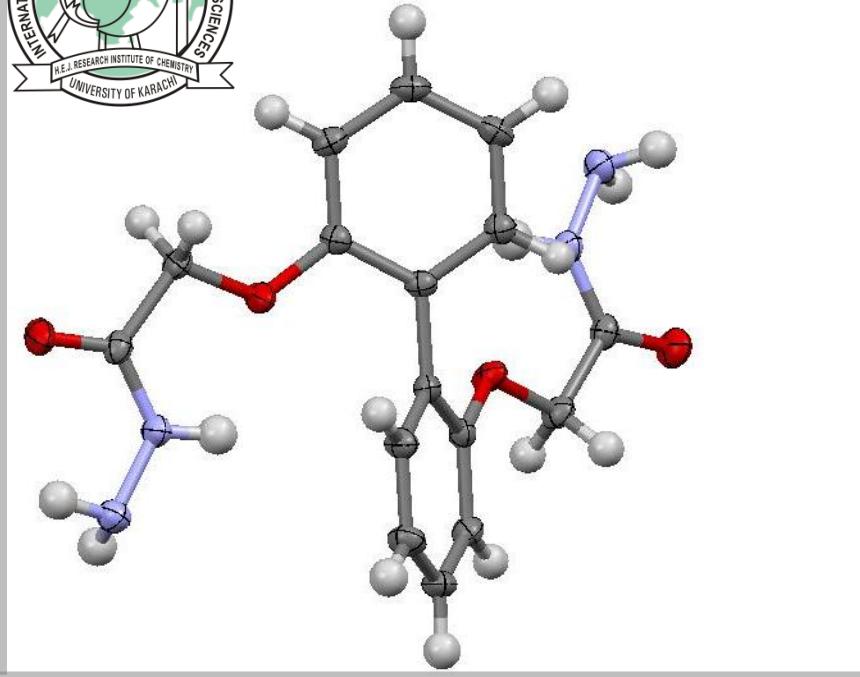
***t*-BLM formation and characterization of Supramolecular Nanopores via AFM**



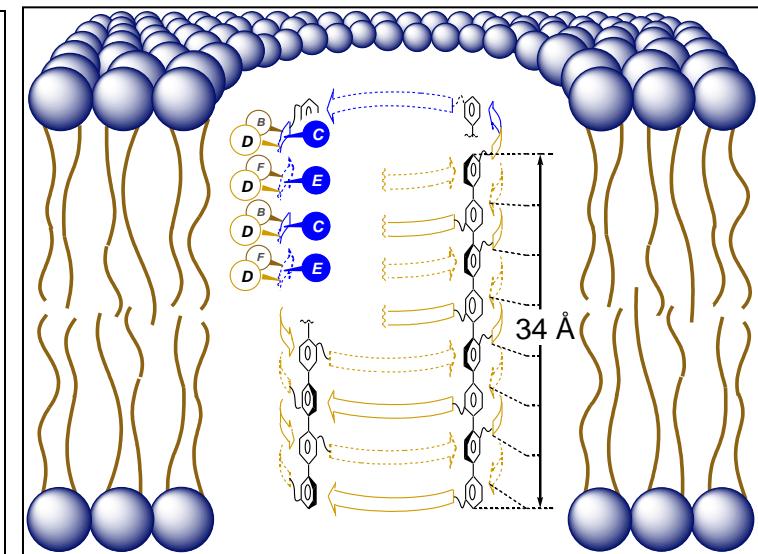
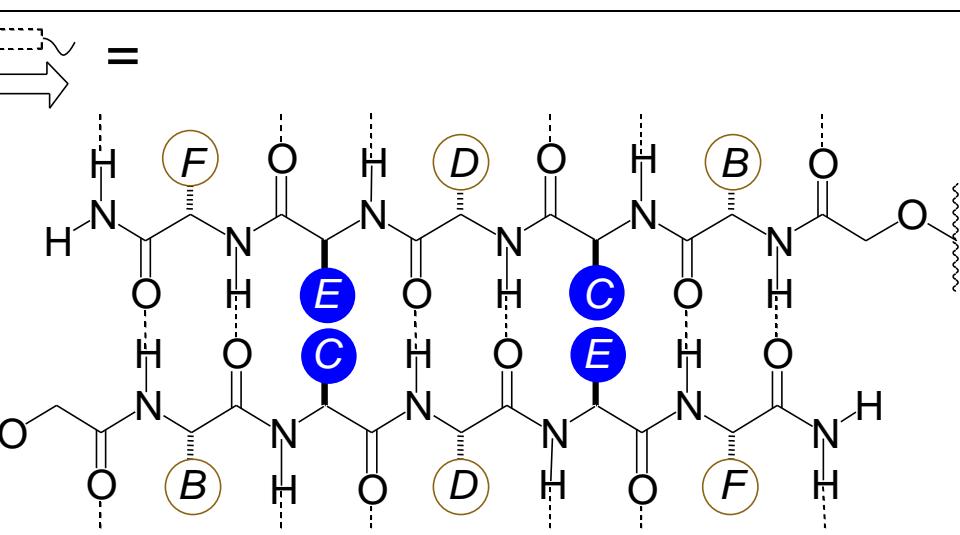
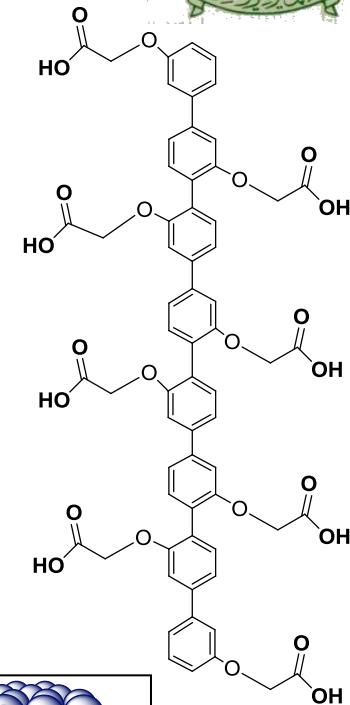
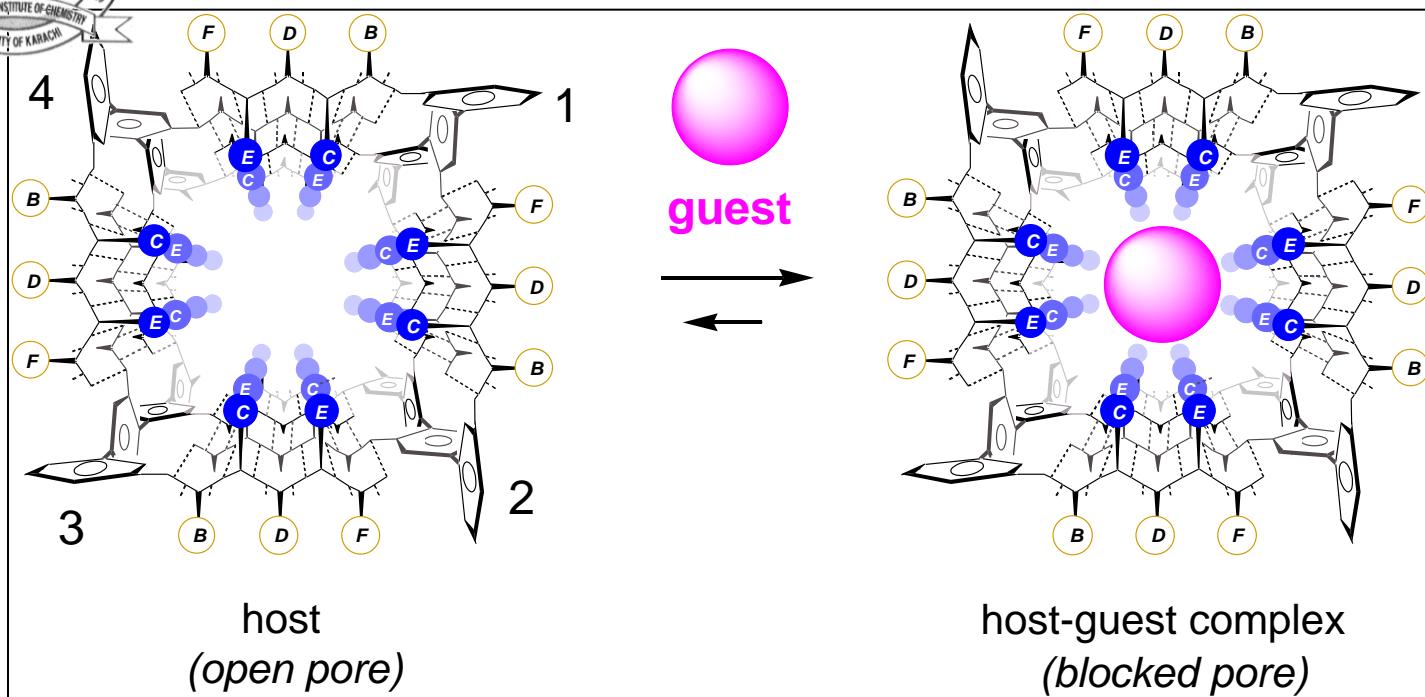
Applications of Ion Channel

- 1) Cell membrane
- 2) Main target class in drug discovery and delivery
- 3) Protein Based Polymer in the form of Nanopores are active areas for Drug Delivery.
- 4) 50% currently available drugs interact with cell membranes through Nanopores, well known examples are Nicotine, and Inositol trisphosphate etc.
- 5) Despite the physiological significance and therapeutic relevance in a wide variety of biological systems, ion channels still remain under exploited as drug targets. This is to a large extent resulting from the historical lack of screening technologies to provide the throughput and quality of data required to support medicinal chemistry.

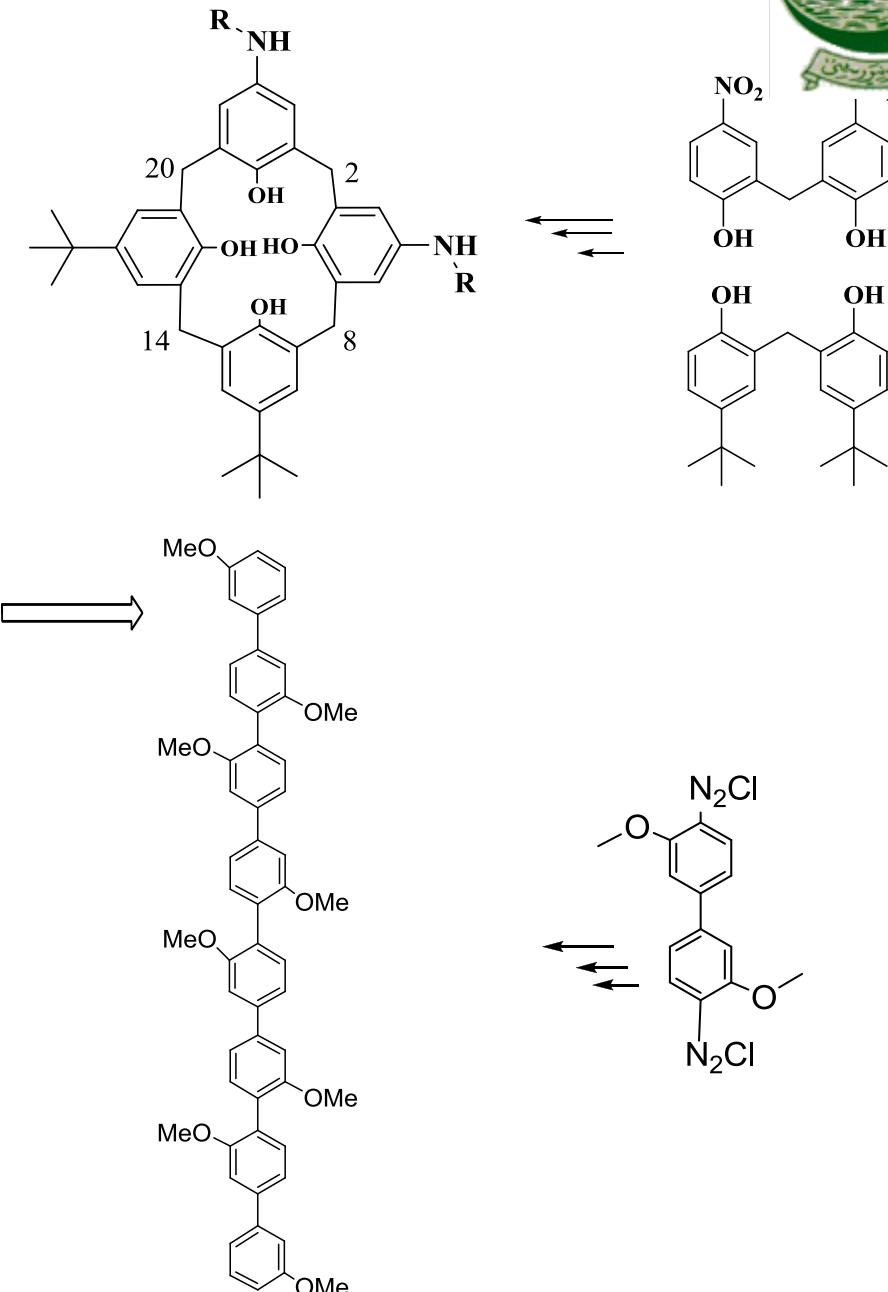
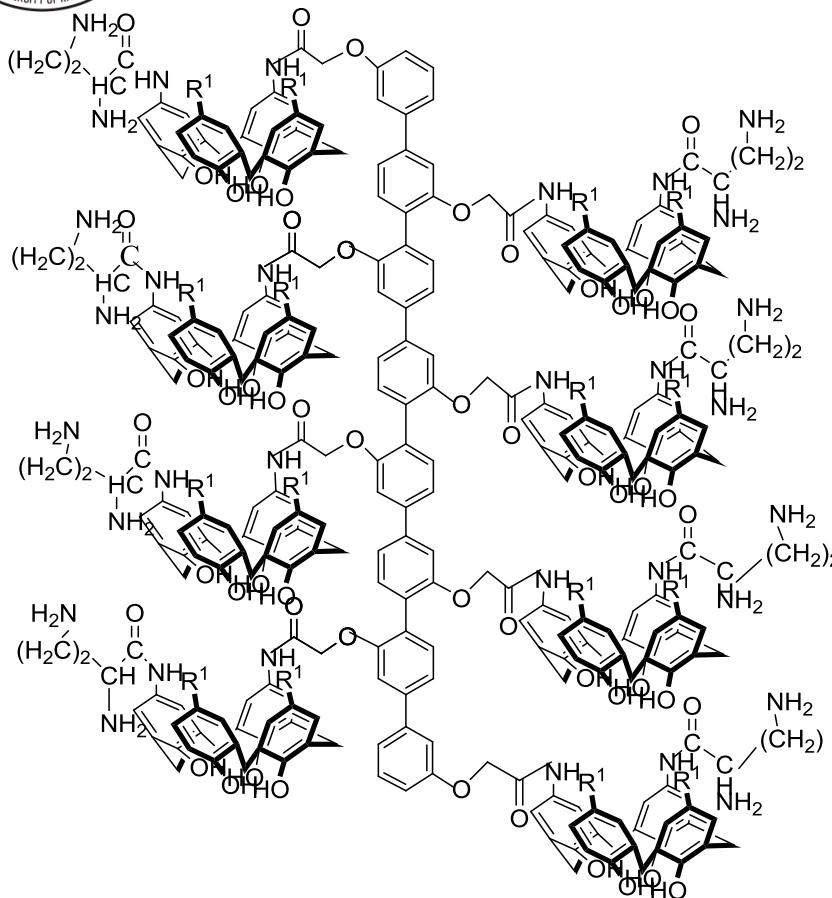
Nanopores via Macrocycles



Synthetic ion channels and pores with Rigid-Rod molecules



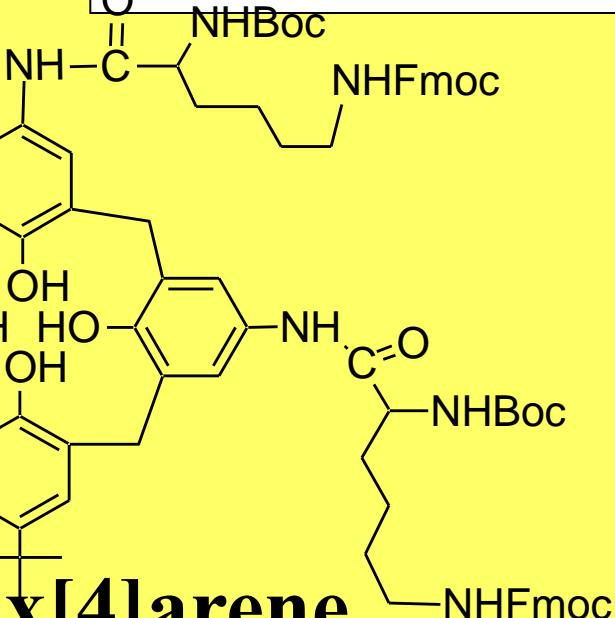
Synthesis of *p*-Octiphenyloctacalix[4]arene



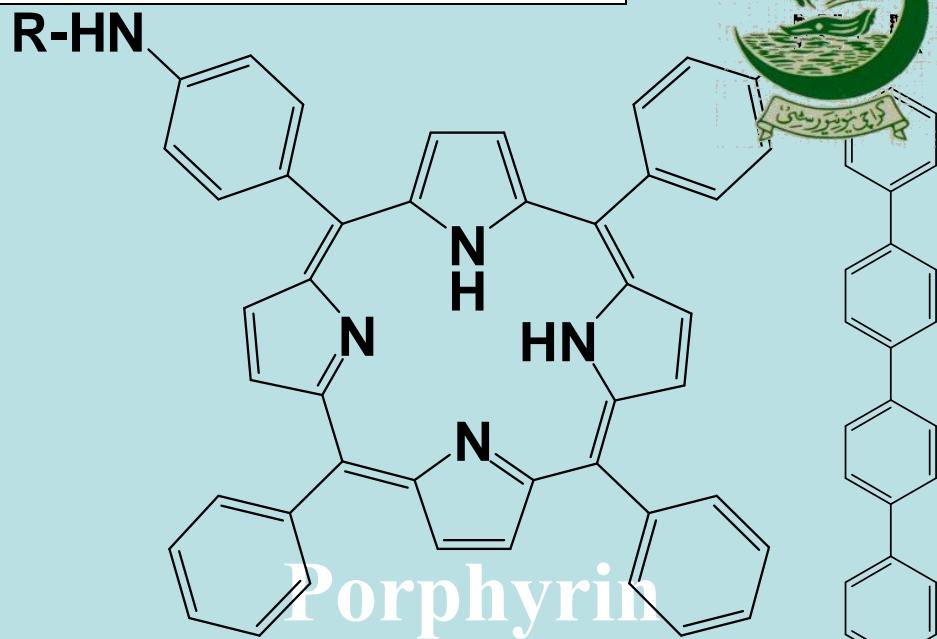
***p*-Octiphenyloctacalix[4]arene
Multifunctional Pore**



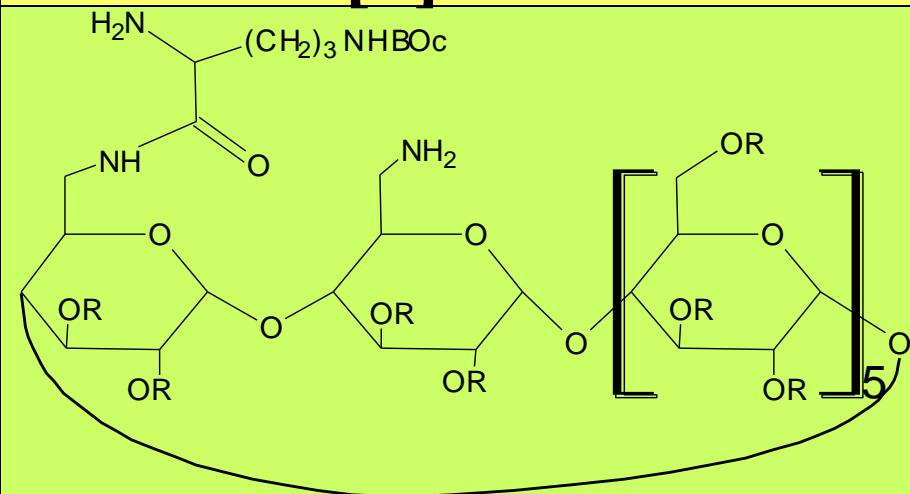
Hoop and Stave for the Formation of Barrel



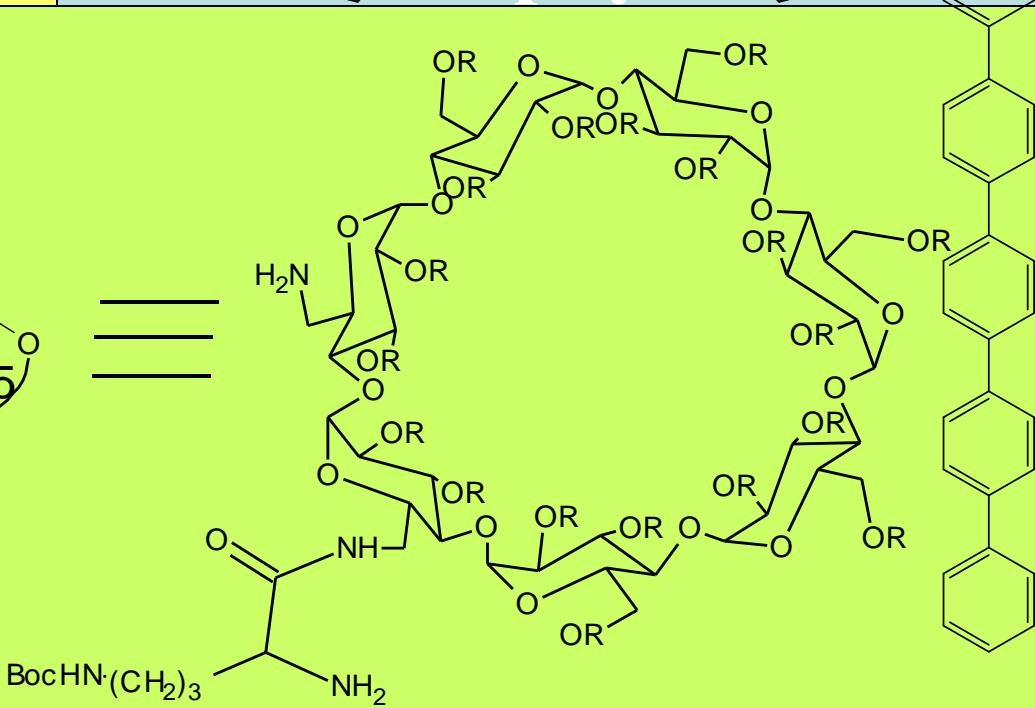
Calix[4]arene



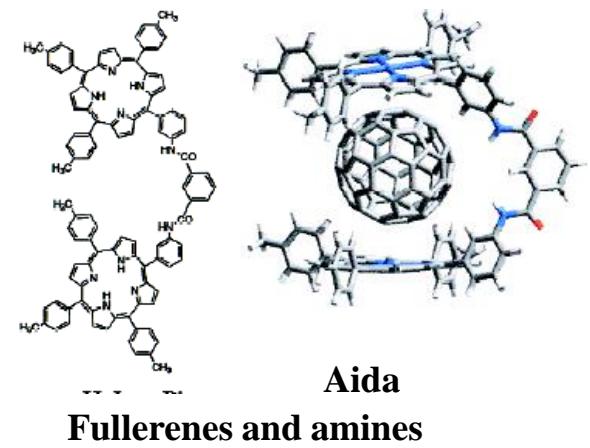
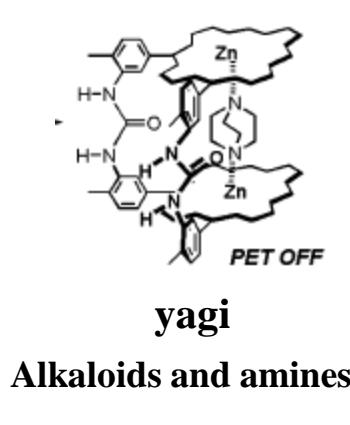
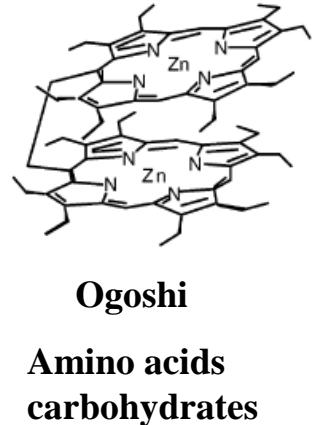
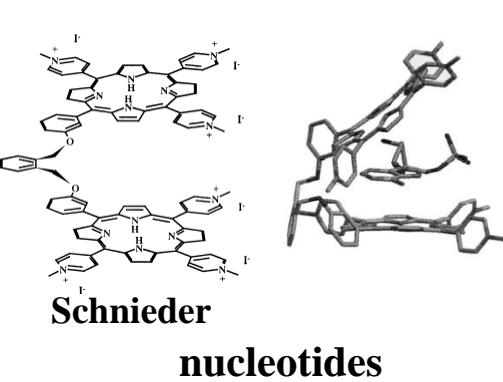
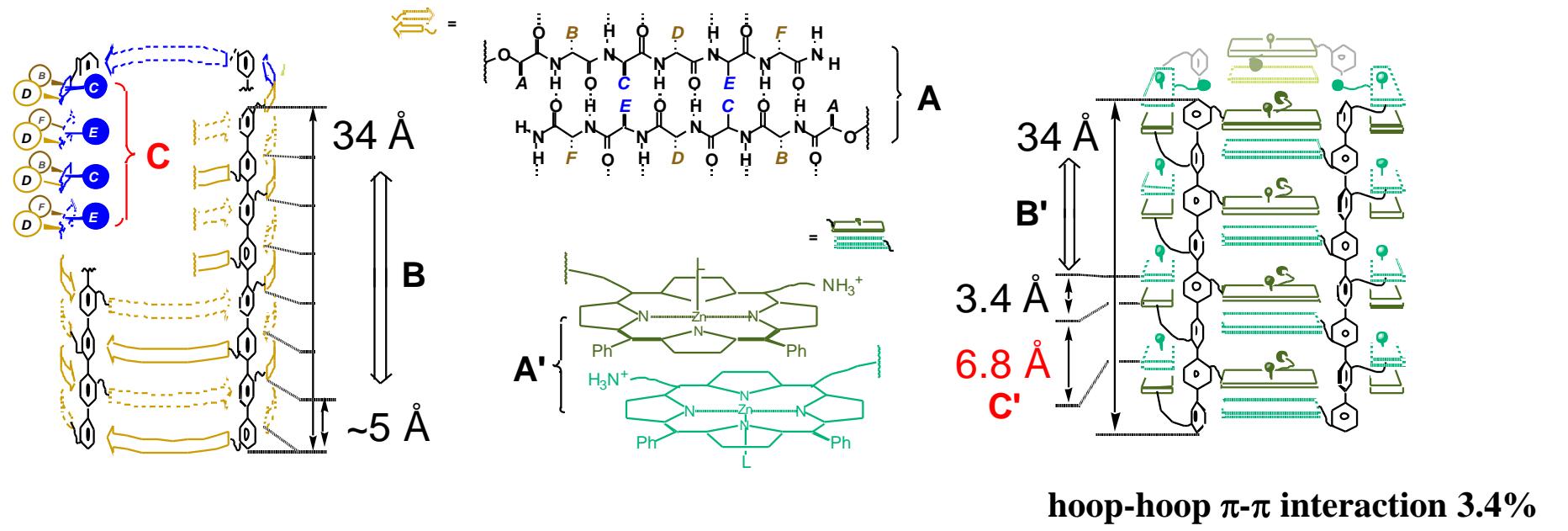
Porphyrin



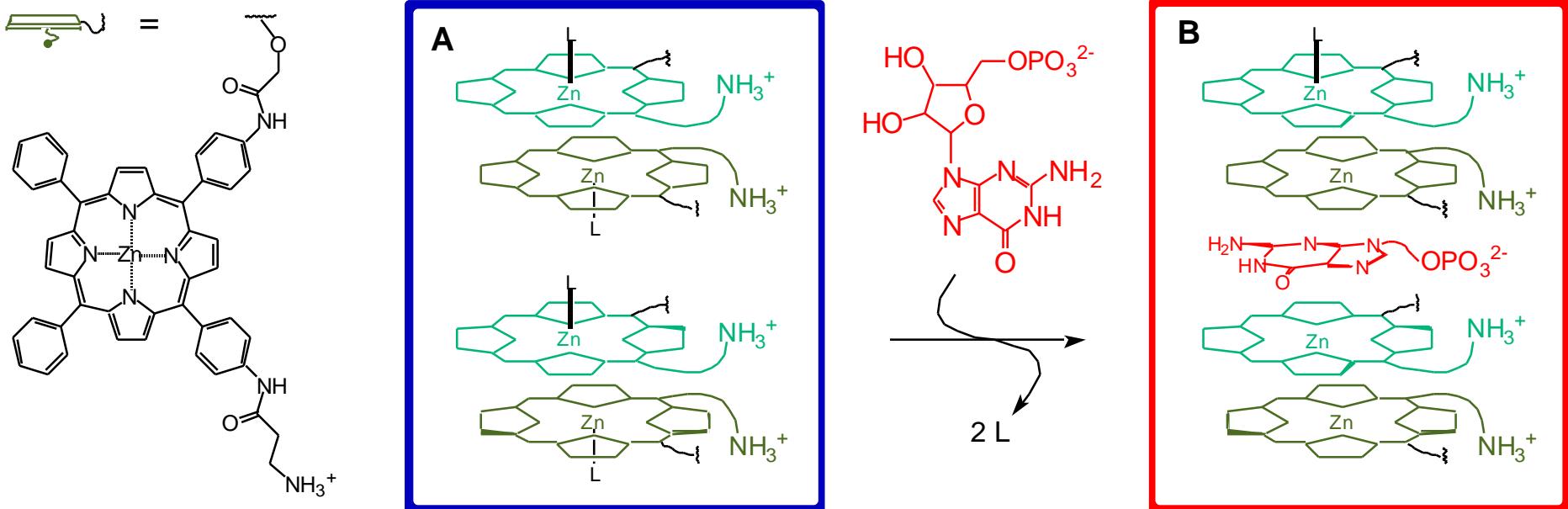
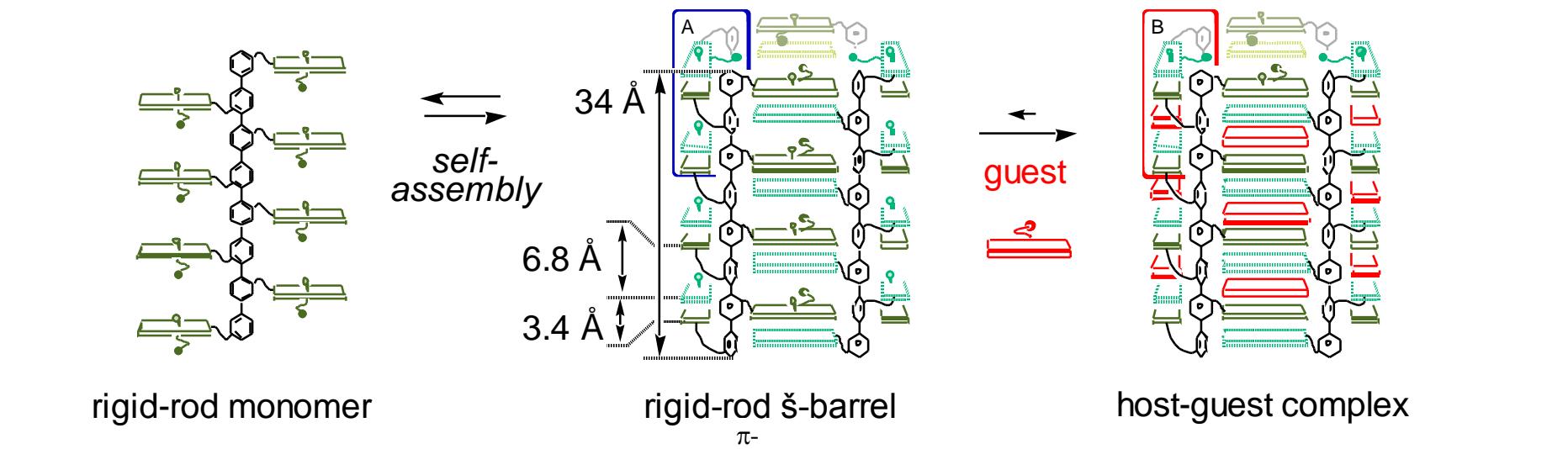
Cyclodextrin



New Synthetic Multifunctional Pores: From Rigid-Rod β -Barrels Toward Rigid-Rod π -Barrels

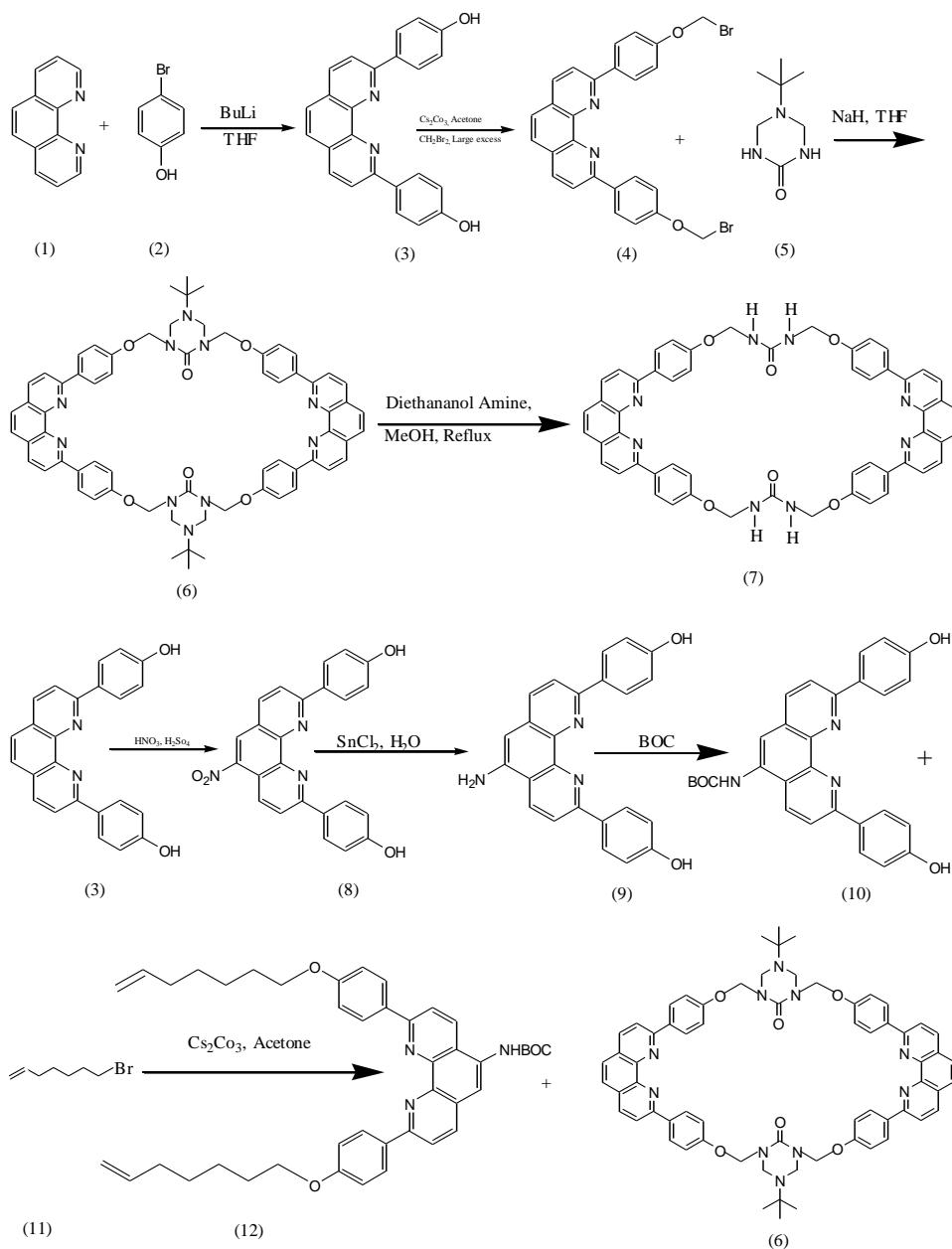


Design of Rigid-Rod π -Barrels as Multifunctional Pores (Specific Examples)





Pores via Urea base macrocycles



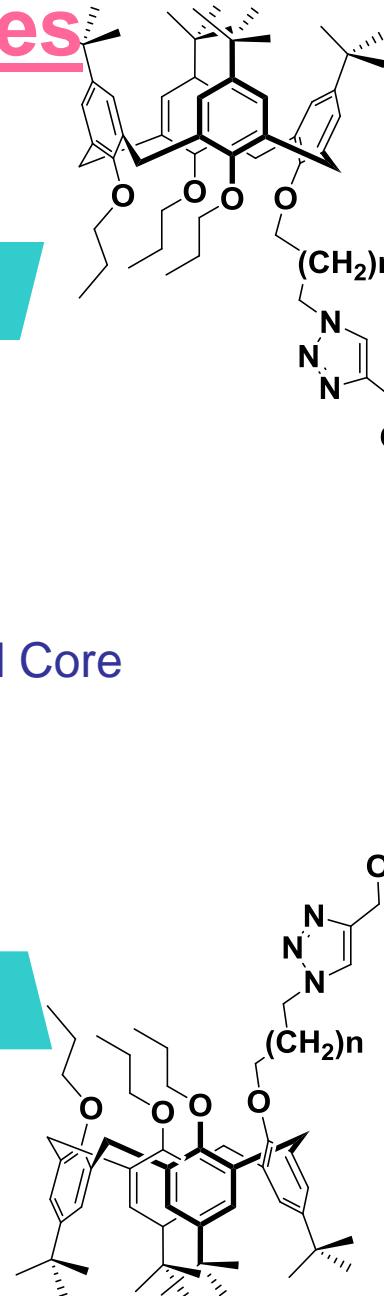
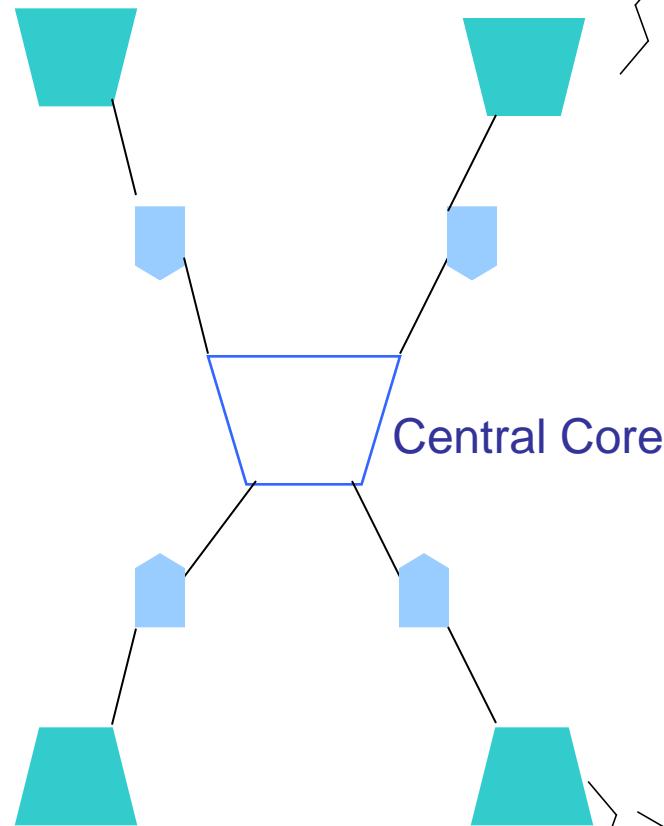
Hydrogen Bonding and Pi-pi-stacking ability can be exploited to form nanopores

Structural Features

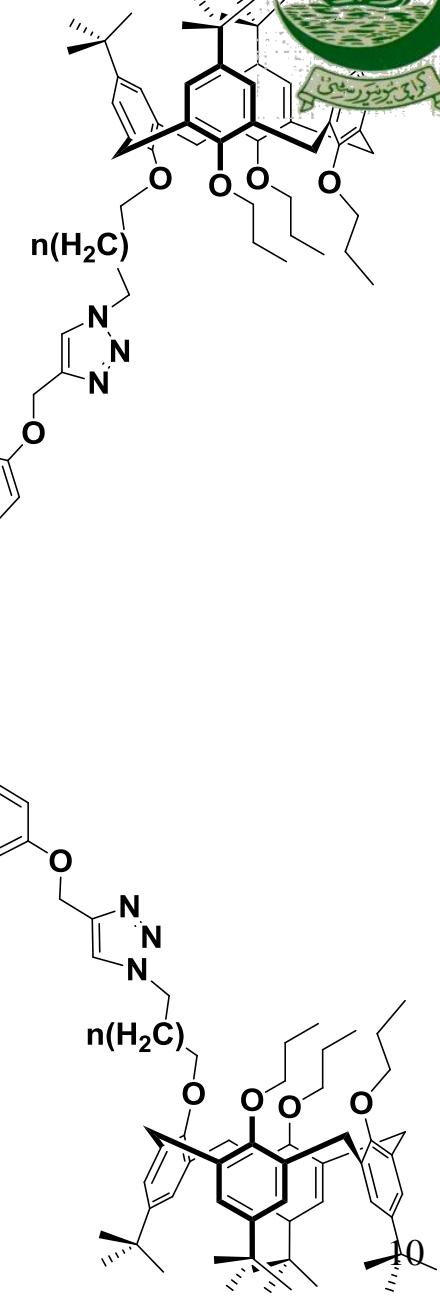
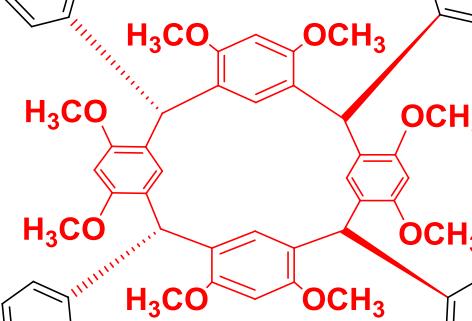
Pores via Hybrid Multicalixarene



Peripheral Units



$$\begin{array}{ll} n=0 & \mathbf{36} \\ 2 & \mathbf{35} \\ 4 & \mathbf{34} \end{array}$$

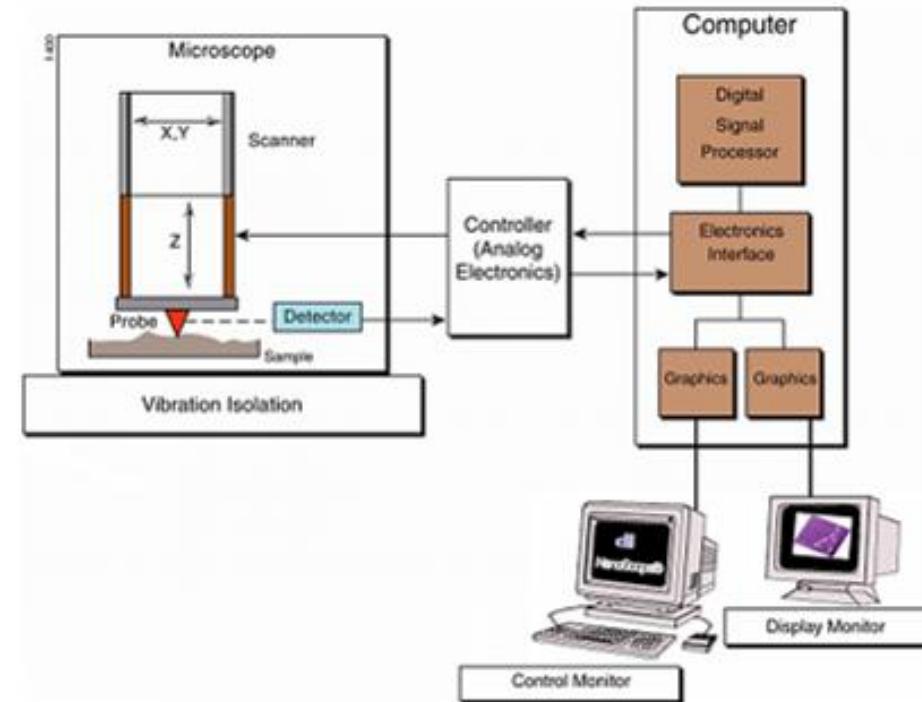


Introduction of AFM

AFM was developed in 1986 by G. Binning and H. Rohrer

A normal AFM instrument consist of the following parts

- ④ probe (cantilevers)
- ④ cantilever deflection monitoring system (The AFM head uses a beam deflection system to monitor the cantilever displacement.)
- ④ scanning system (Piezoelectric scanner)
- ④ electronics
- ④ computer

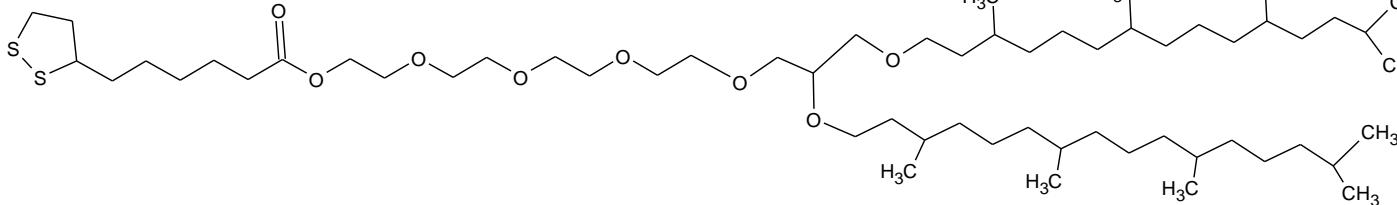




Molecules Used for SAM Formation

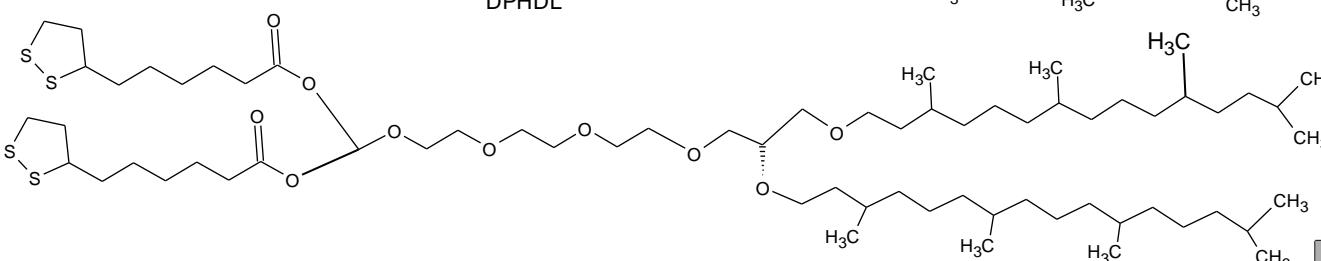
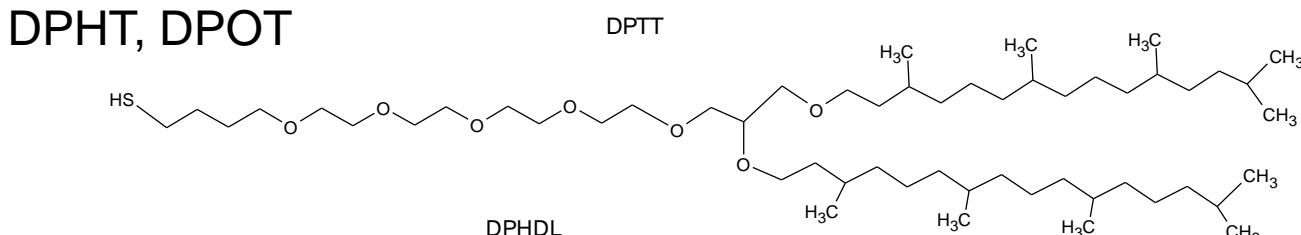


DPHL
DPOL
DPTDL

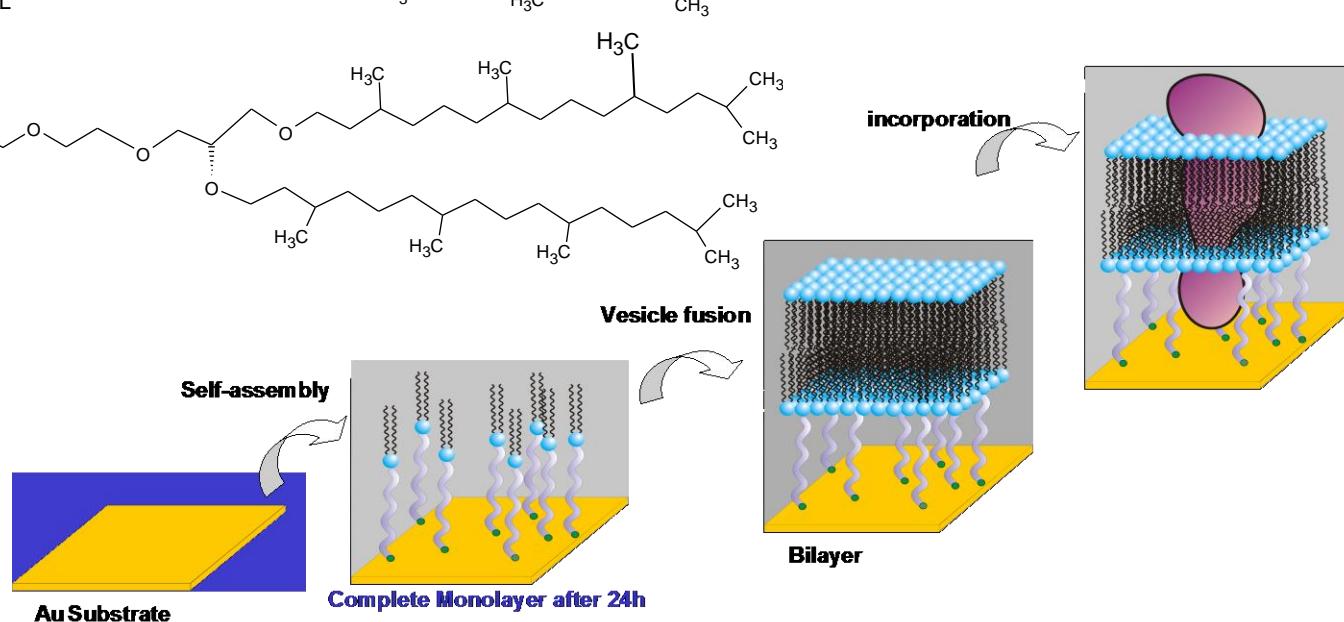


2,3-di-O-phytanyl-sn-glycerol-1-tetraethylene glycol-D,L-lipoic acid ester lipid (DPTL).

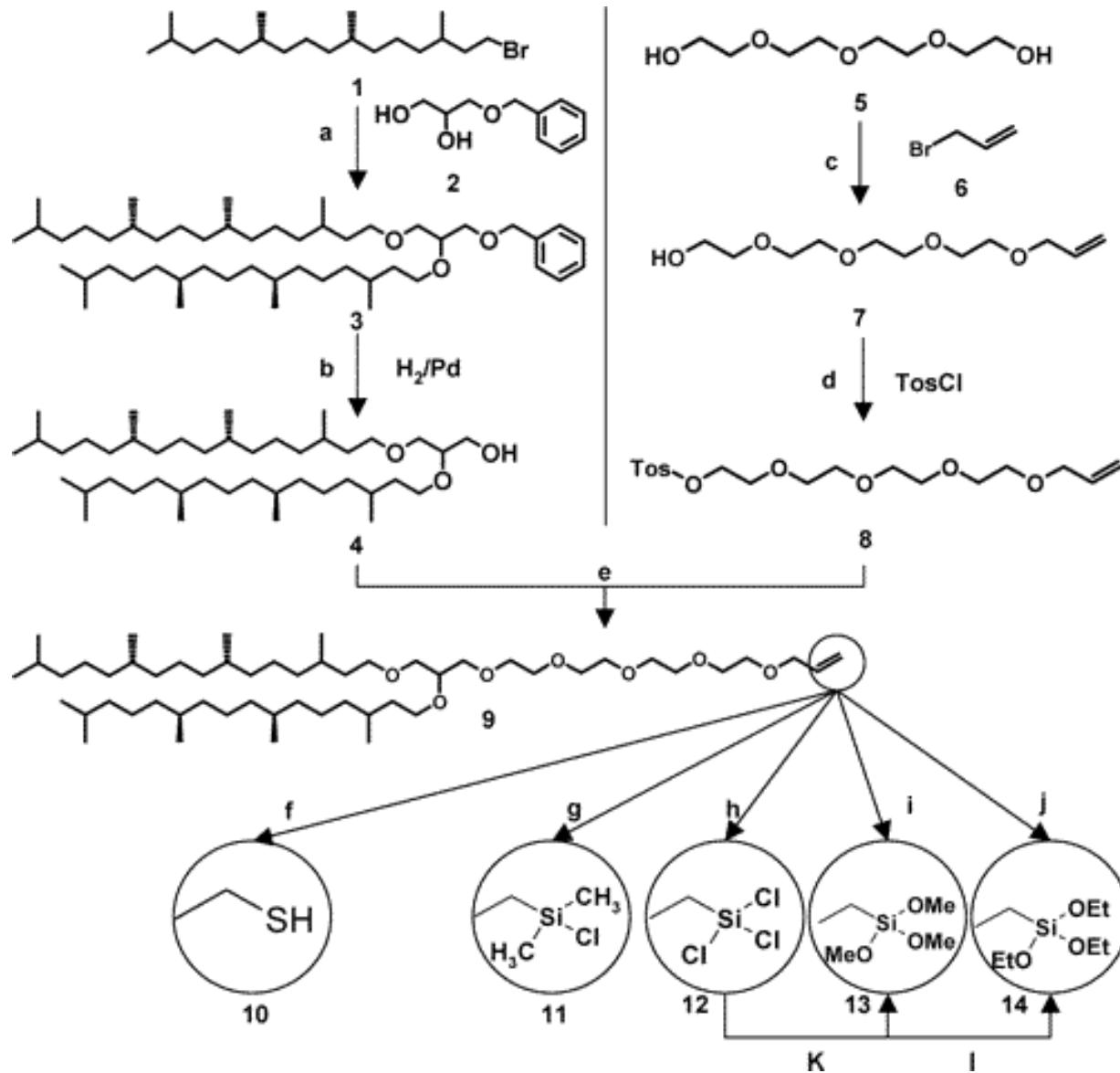
DPHT, DPOT



Investigation step by step tBLM



Synthesis of Lipids





Problems in resolution and tip

- 1) Cantilevers silicon 70 kHz frequency (Soft, produce high resolution)
- 2) Cantilevers kept under vacuum for overnight
- 3) Cantilevers plasma cleaned prior to use
- 4) Liquid cell along with O-ring sonicated in detergent for 30 minutes at 40°C
- 5) Sonicated in ethanol for 30 minutes
- 6) Rinsed extensively with Milli-Q
- 7) Everything dried with flush of nitrogen
- 8) Extremely soft taping
- 9) Target amplitude was kept in between 2-3.
- 10) For higher resolution always Z-range was decreased.

Proposed Mechanism for the Self-assembly Process

DPTL

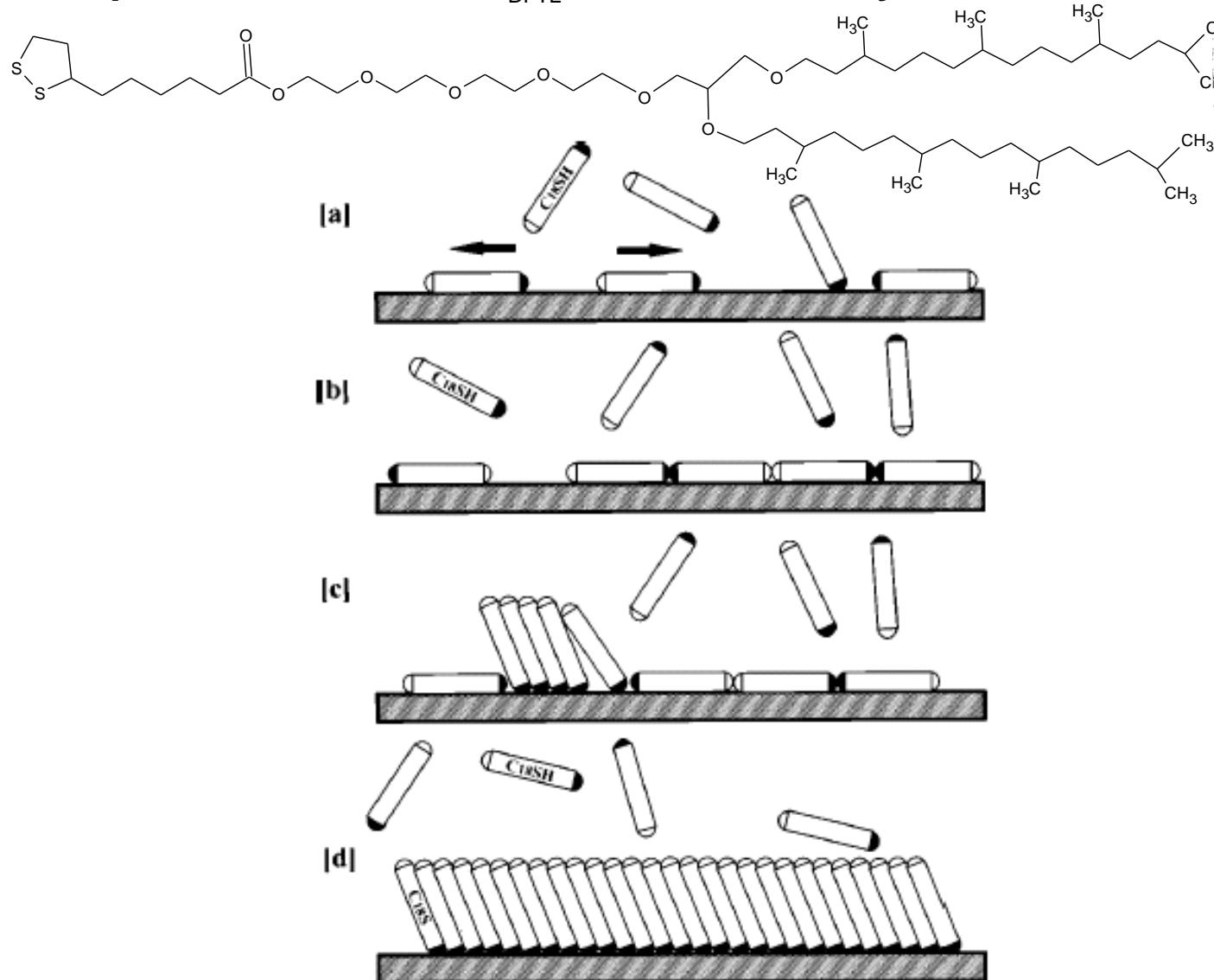
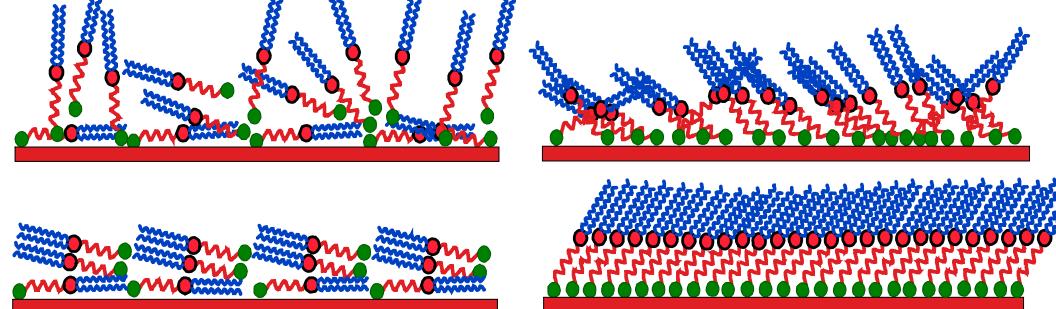
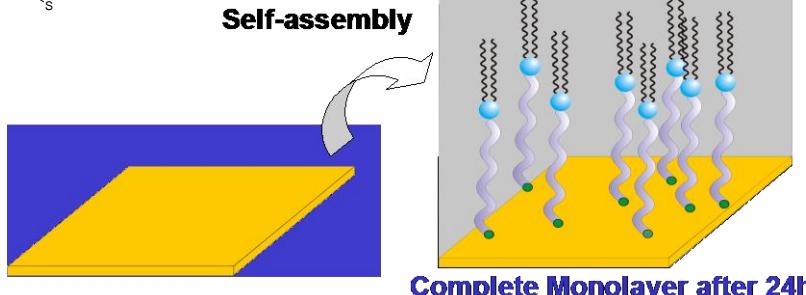
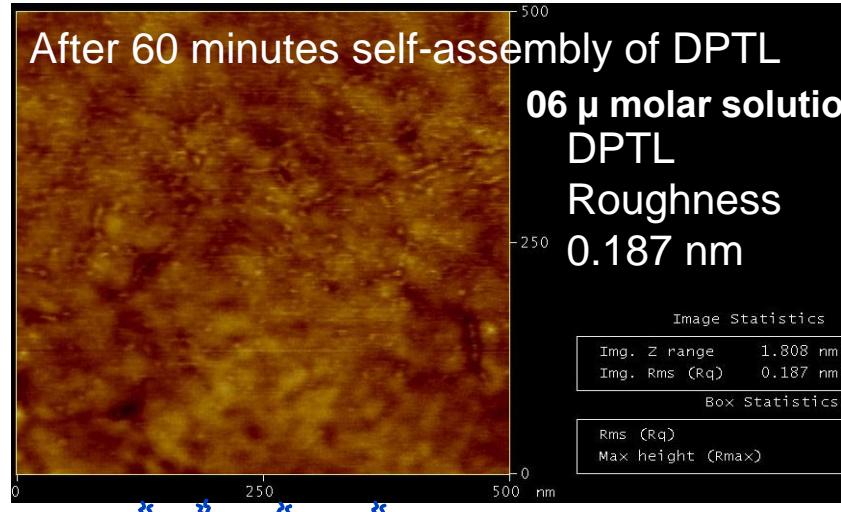
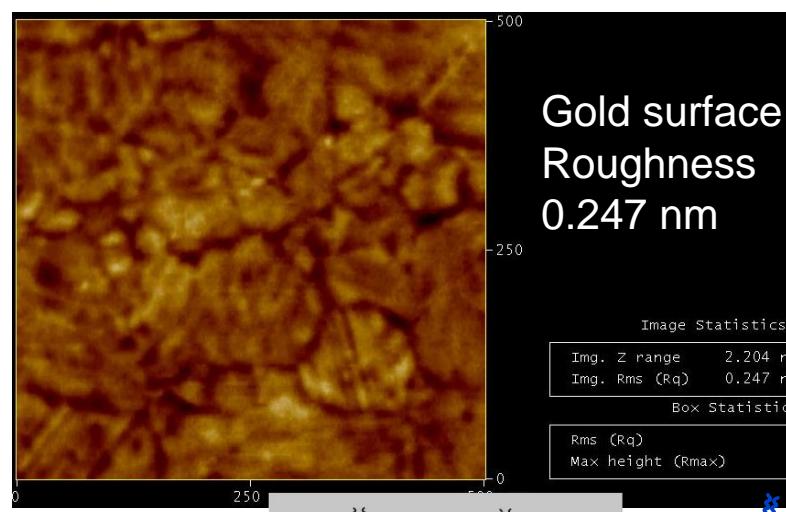
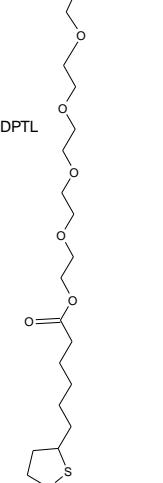
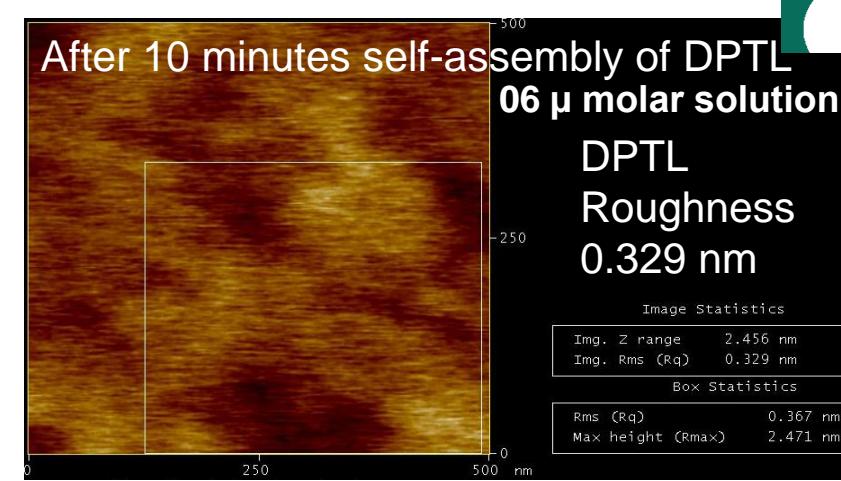
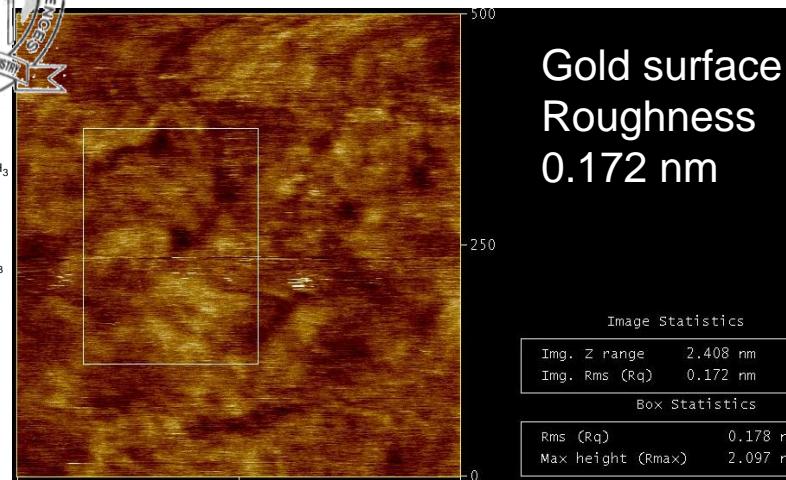
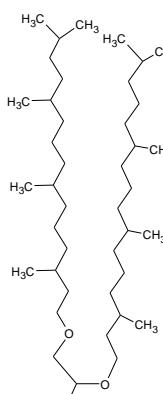


FIG. 5. Schematic mechanistic diagram for the self-assembly of $\text{CH}_3(\text{CH}_2)_{17}\text{SH}$ on $\text{Au}(111)$. (a) Initial adsorption, (b) lying-down phase, (c) Two-dimensional phase transition from a lying-down to a standing-up configuration, (d) Formation of a complete SAM.

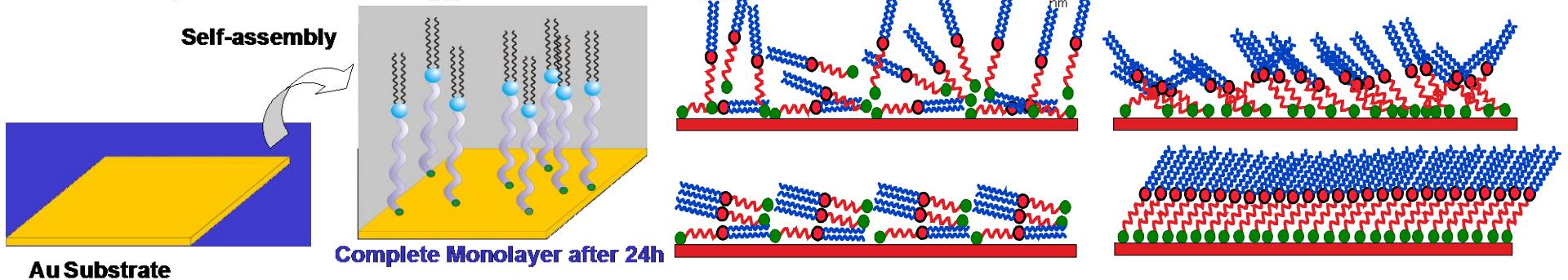
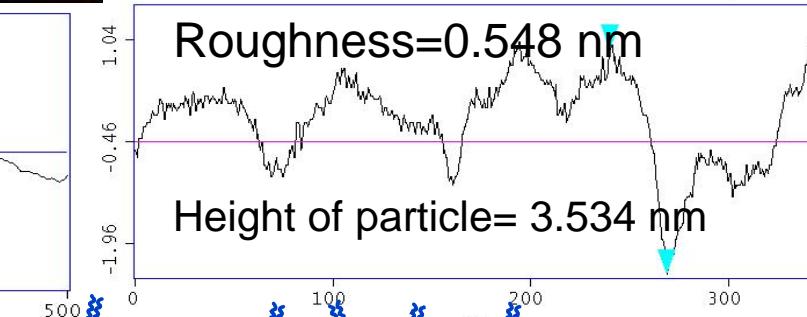
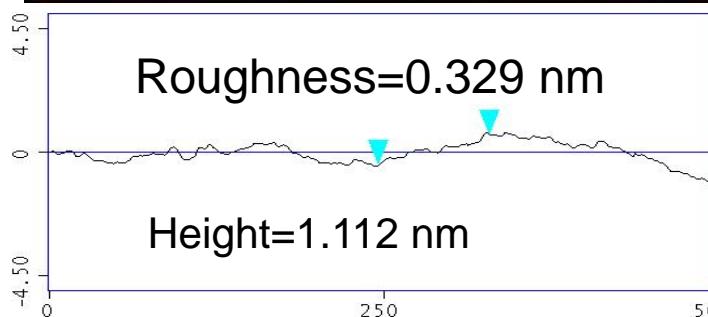
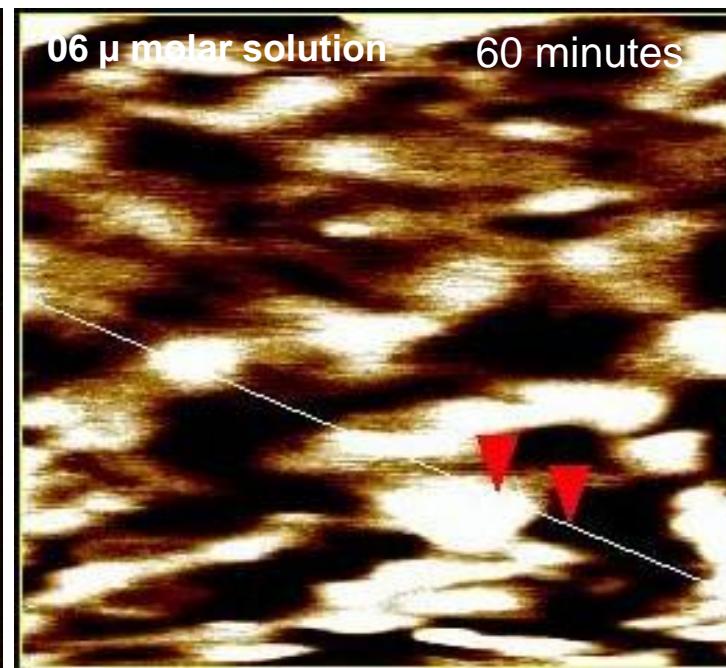
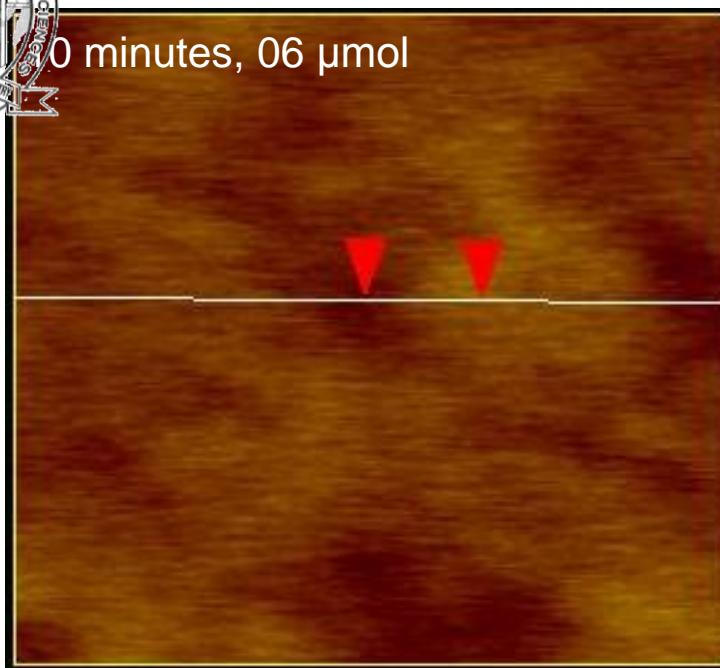
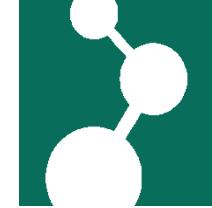


Differences in roughness indicate the self assembly process



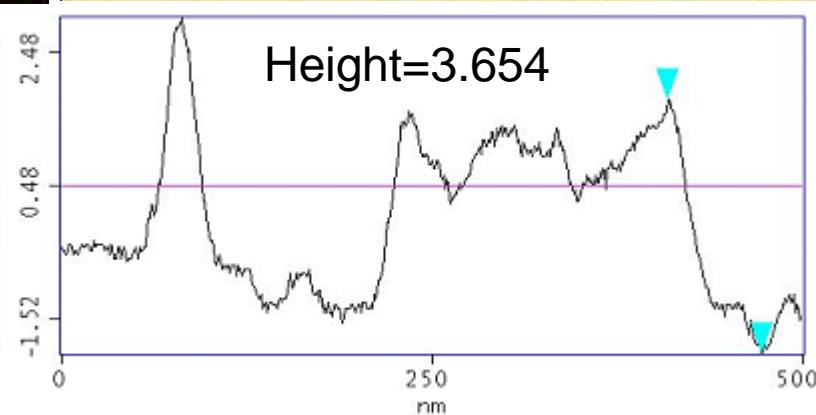
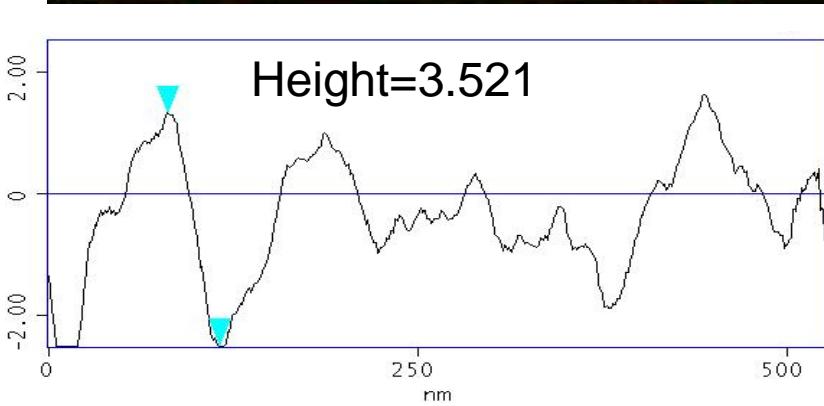
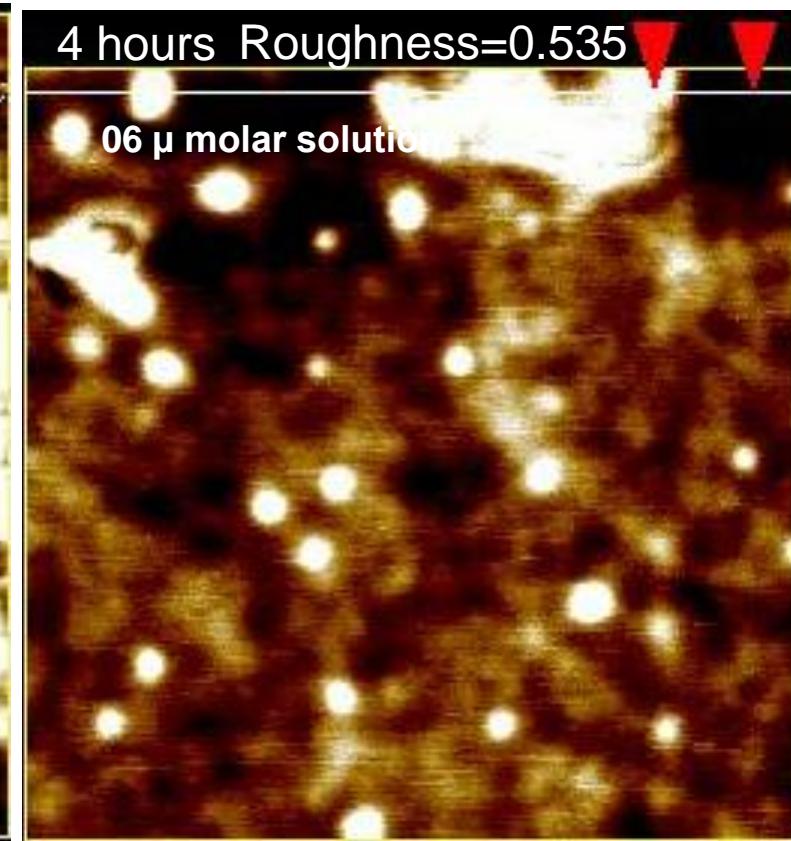
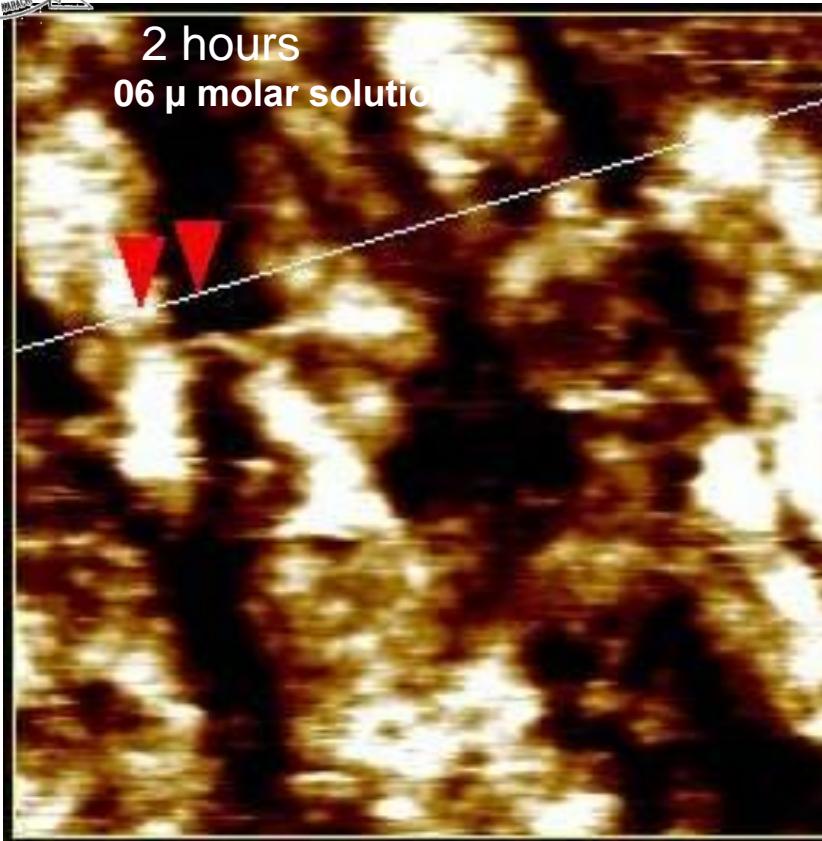


Time dependent cross-sectional analysis of DPTL after 10 and 60 minutes



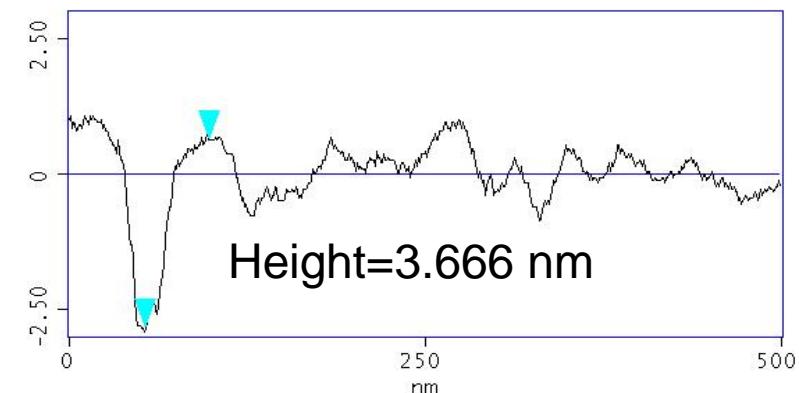
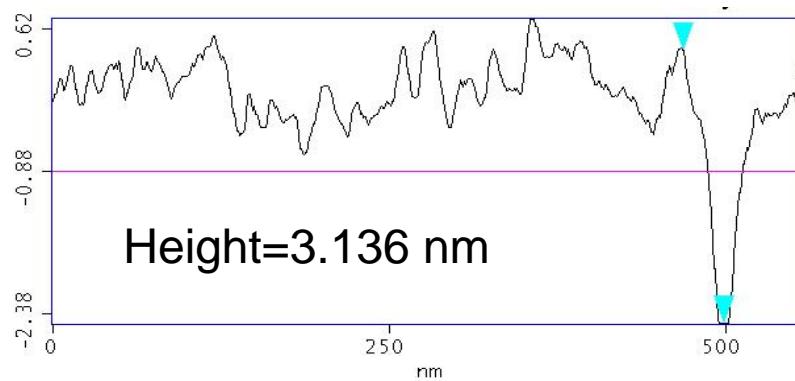
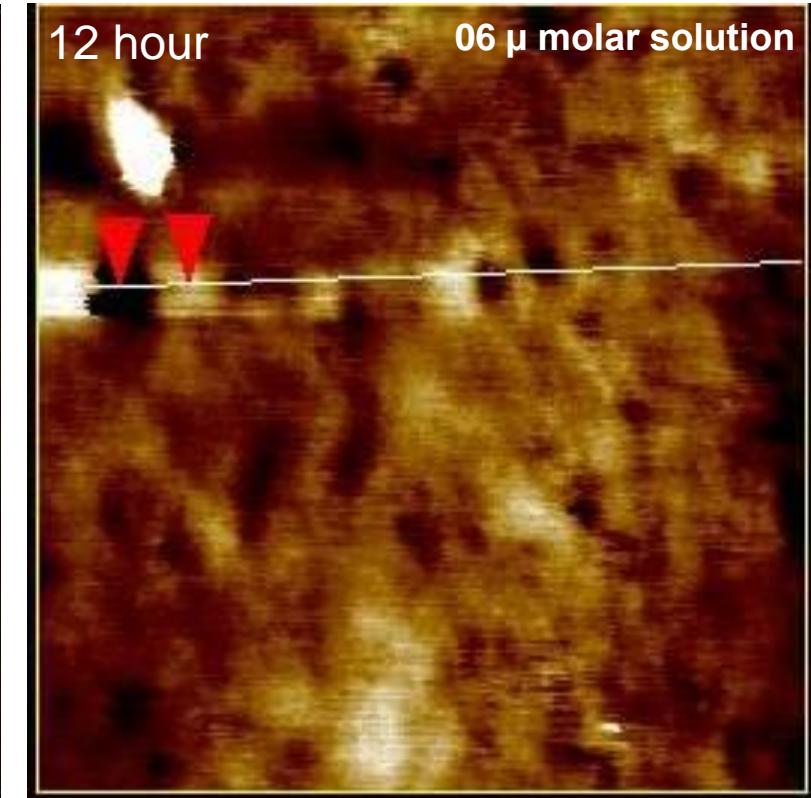
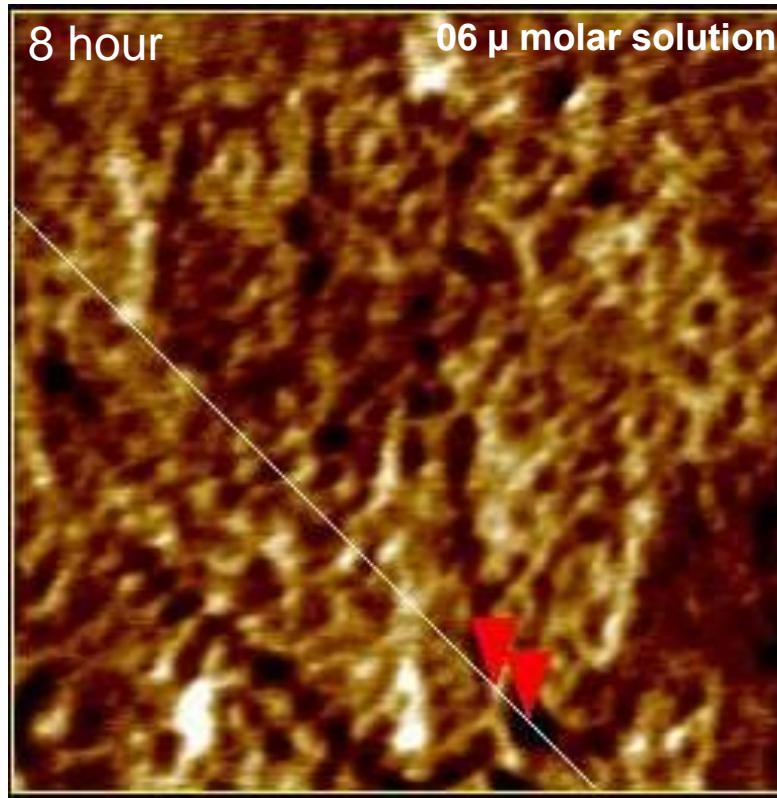


Time dependent cross-sectional analysis of DPTL after 2h and 4h.





Time dependent cross-sectional analysis of DPTL after 8h and 12h



Time dependent cross-sectional analysis of DPTL after 24 hours

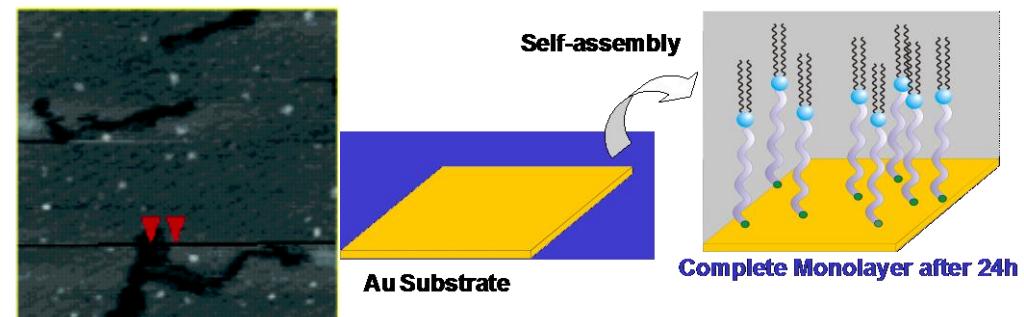
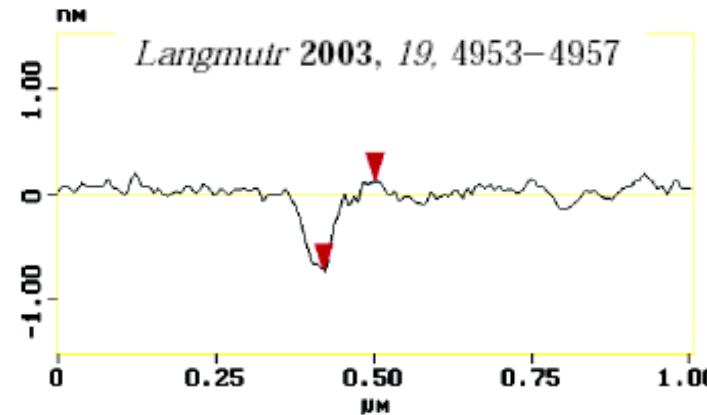
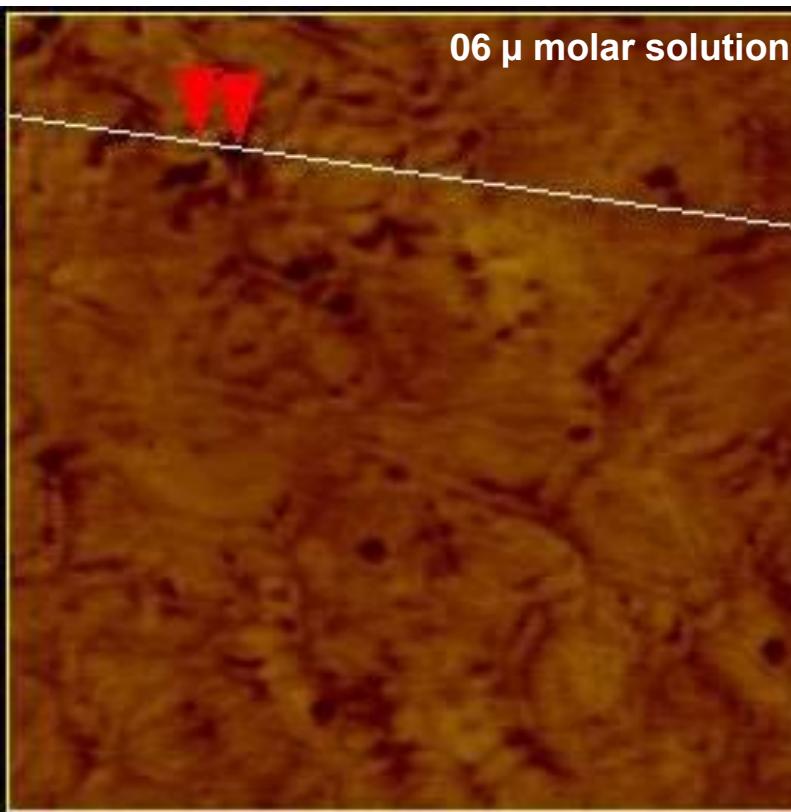
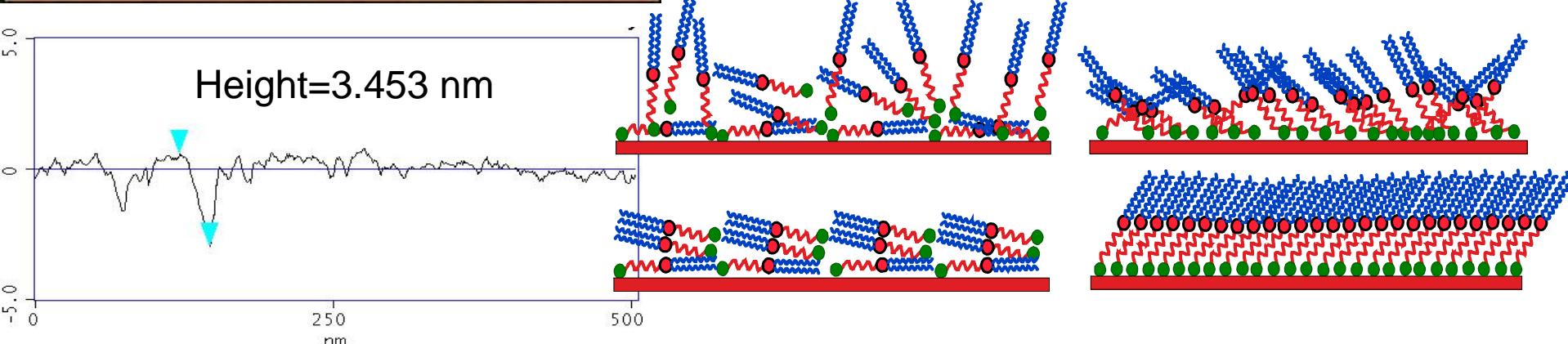


Figure 2. Section analysis of octanol film adsorbed on mica for 24 h. The depth of the two layers was estimated as 0.8 nm.





Lipid bilayer formation via vesicle fusion

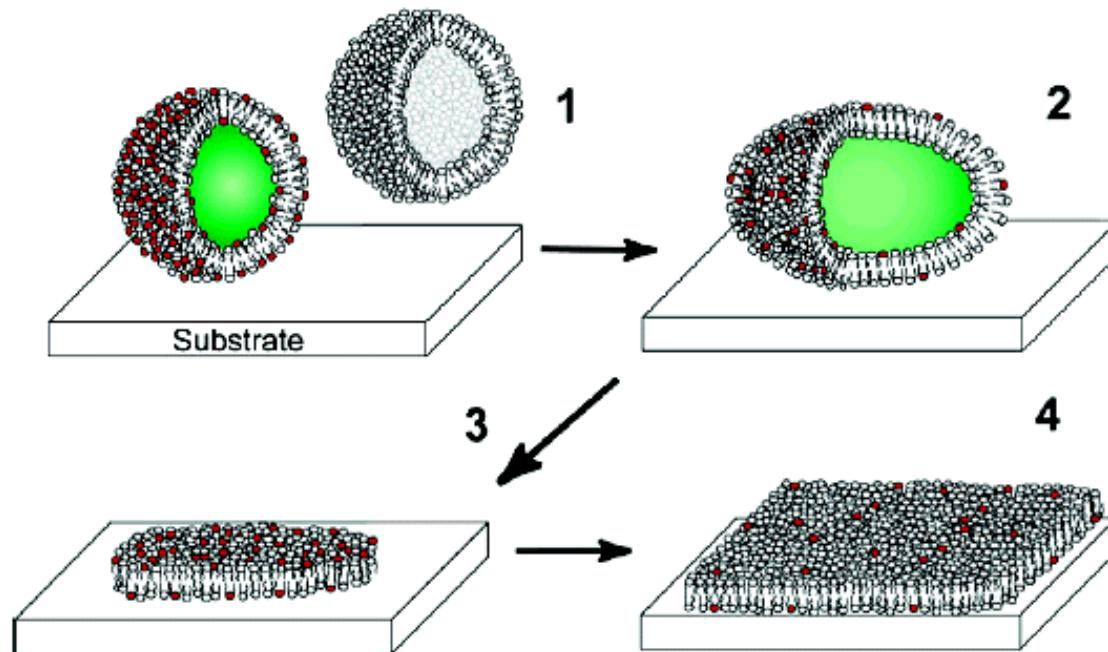
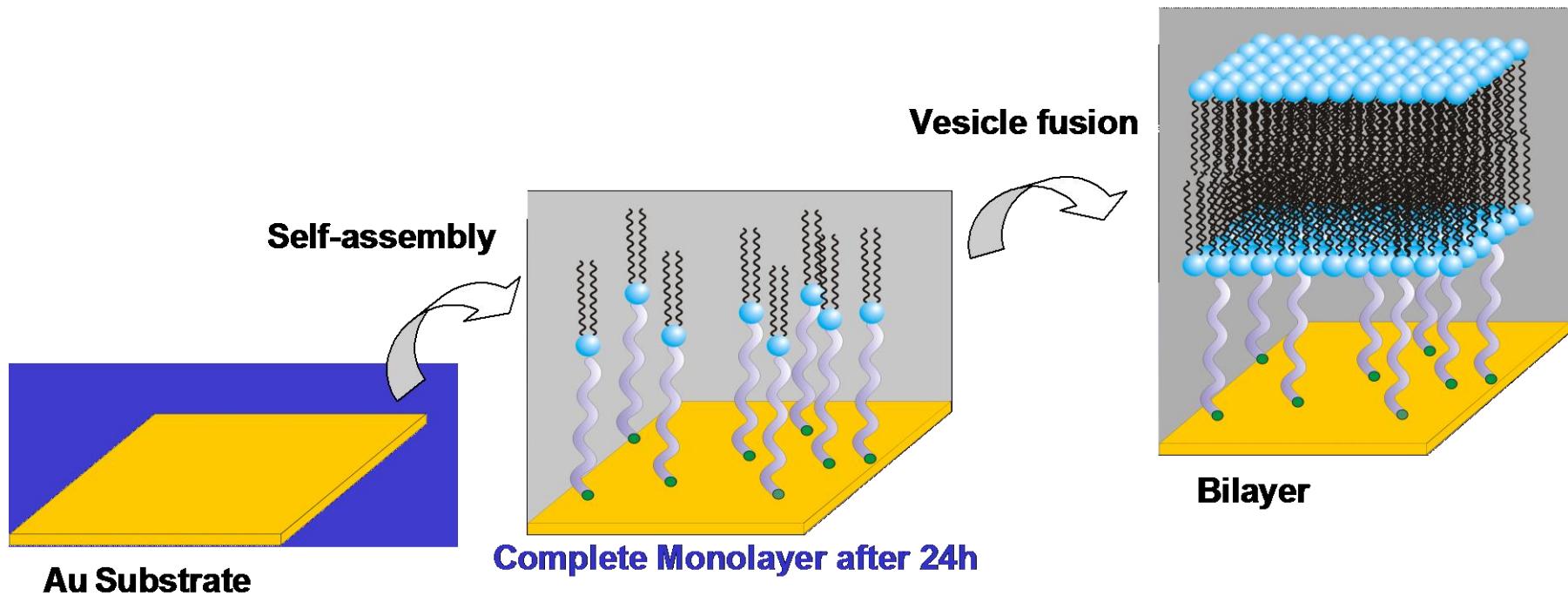
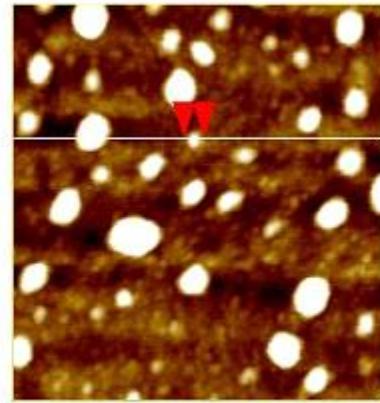
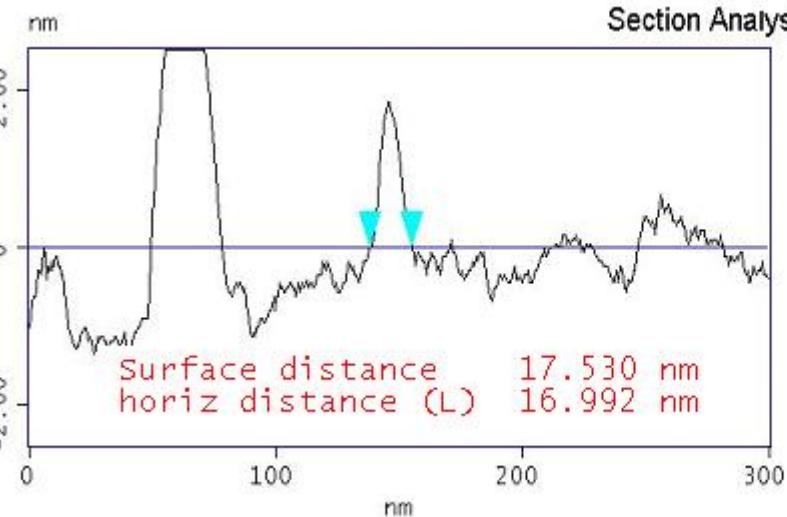
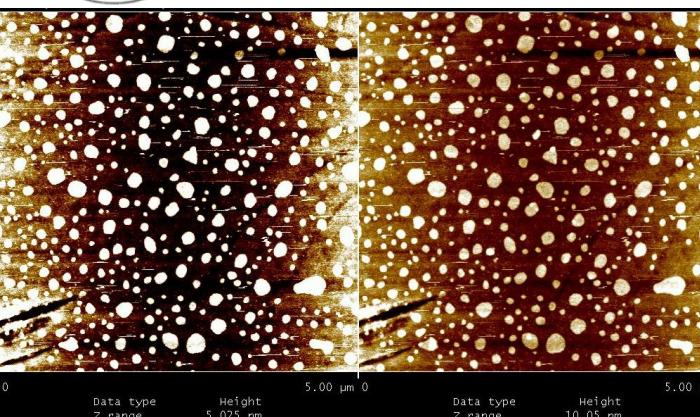
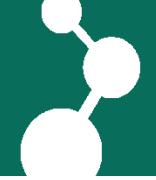


Figure 1. Four step scenario of supported bilayer formation via vesicle fusion comprising (1) vesicle adsorption, (2) fusion of vesicles at the surface to form larger vesicles, (3) rupture of the fused vesicles resulting in bilayer disks, and finally (4) merging of the disks. Here red lipids represent a leaflet fluorophore and green represents an entrapped aqueous dye (ref 1).

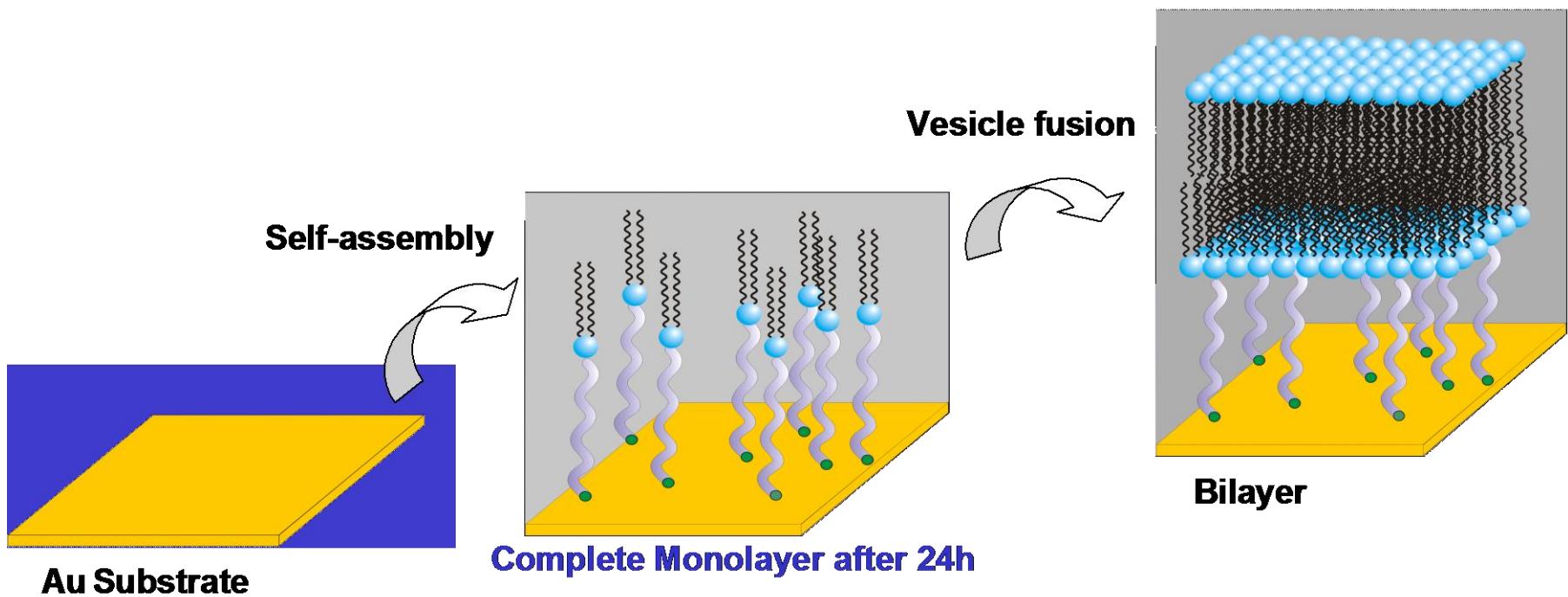
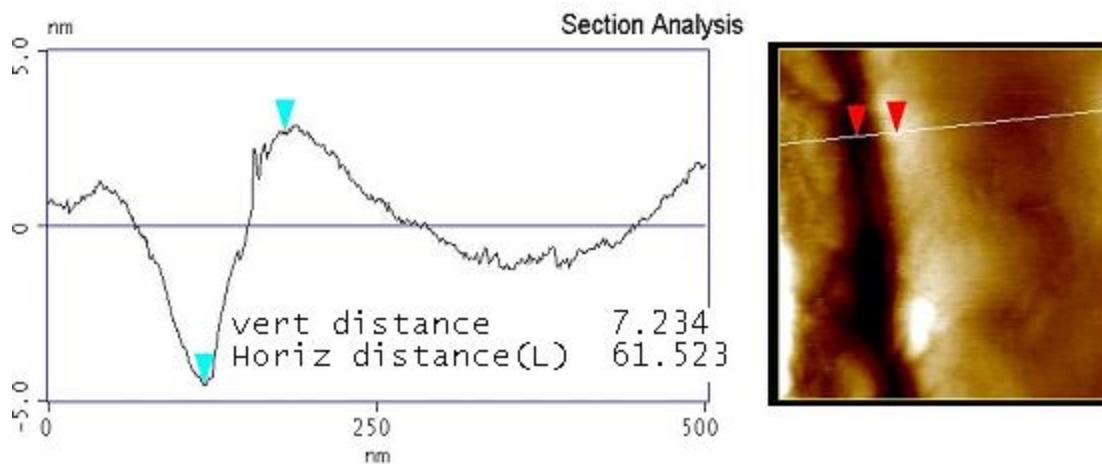
Langmuir 2004, 20, 11600-11606

3 µg/ml solutions of DPhyPC vesicles insitu water 10 min .





50 µg/ml solutions of DPhyPC vesicles. After 10 min of fusion.

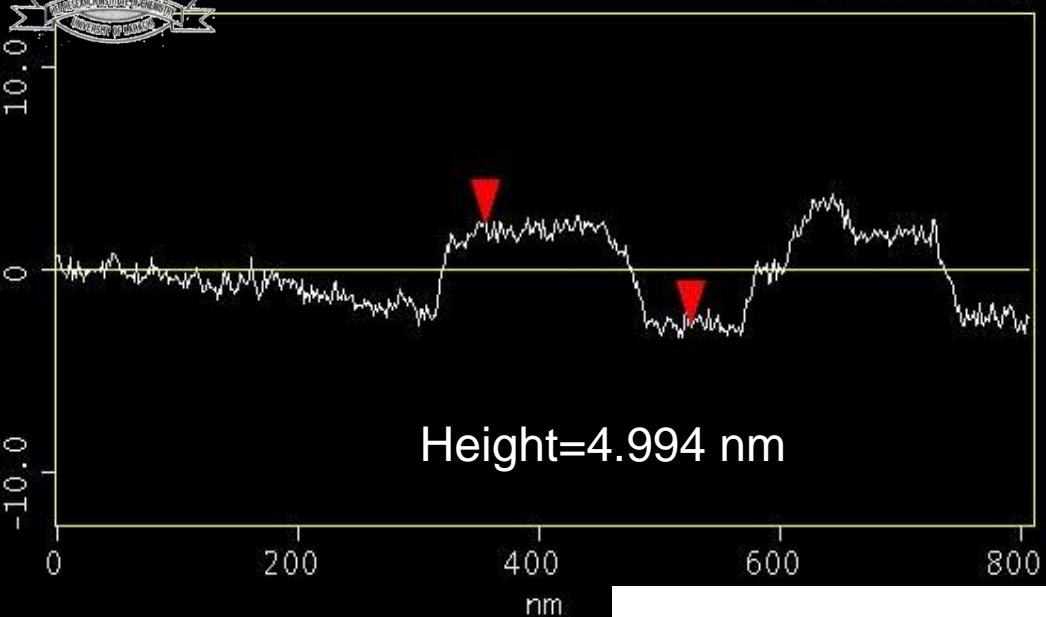




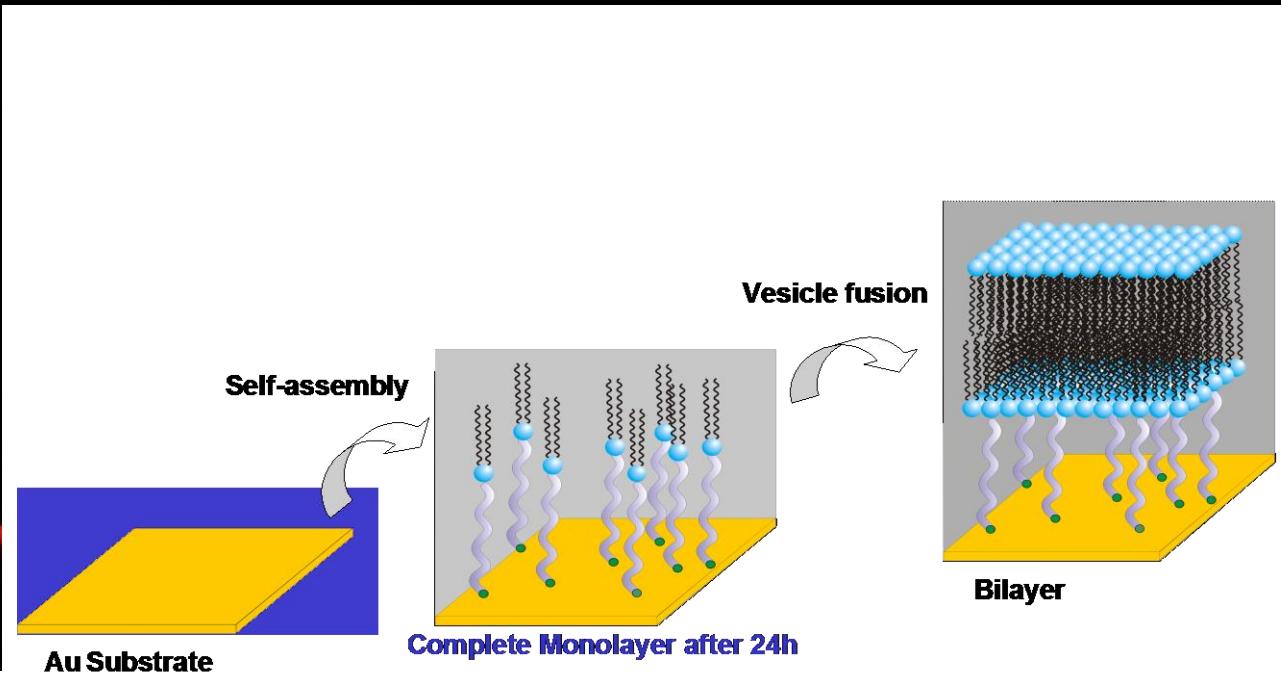
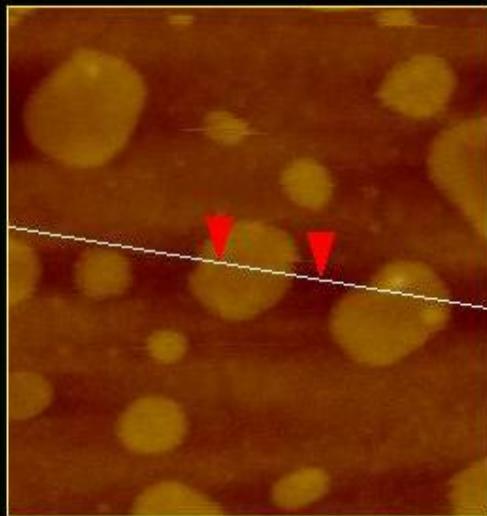
3 µg/ml solutions of DPhyPC vesicles after 10 minutes.



Section Analysis



L	170.31 nm
RMS	1.899 nm
Tc	DC
Ra(Tc)	1.007 nm
Rmax	3.971 nm
Rz	1.625 nm
Rz Cnt	valid
Radius	771.07 nm
Sigma	0.864 nm





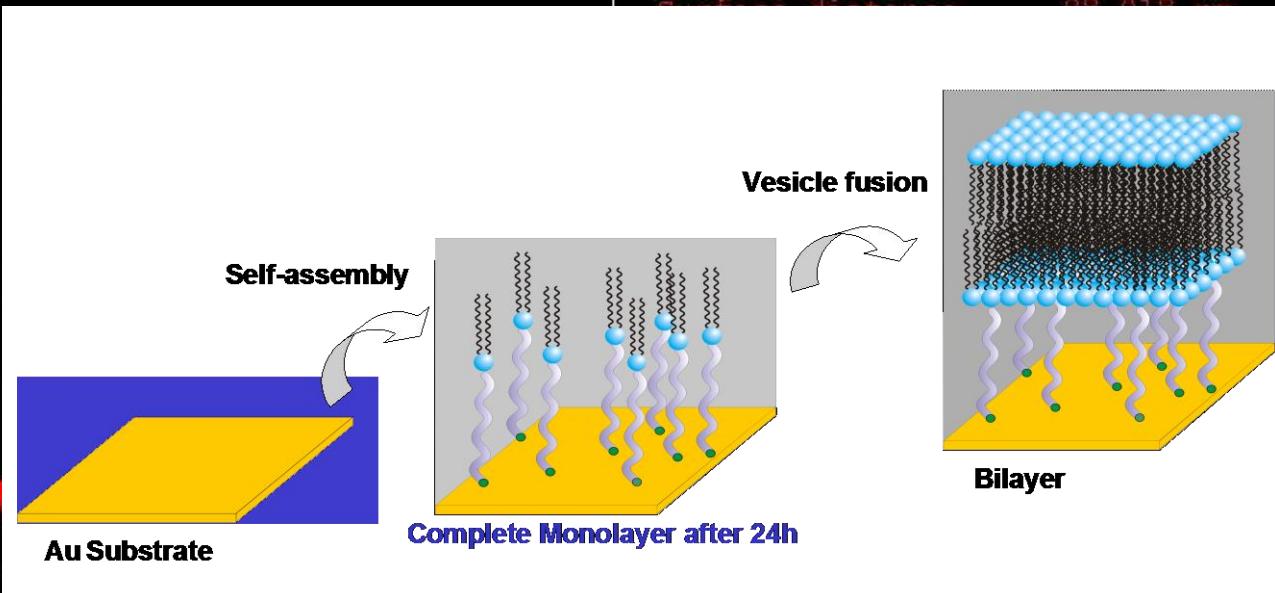
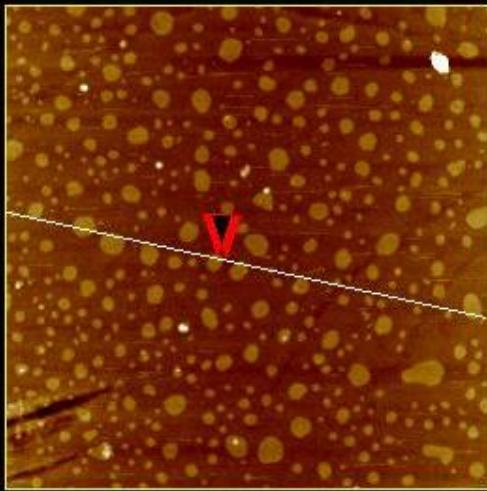
5 μm solutions of DPhyPC vesicles after 10 minutes.



Section Analysis



L	97.656 nm
RMS	1.909 nm
1c	DC
Ra(1c)	0.472 nm
Rmax	2.173 nm
Rz	1.147 nm
Rz Cnt	6
Radius	230.66 nm
Sigma	1.794 nm



3 µg/ml solutions of DPhyPC vesicles after 10 minutes

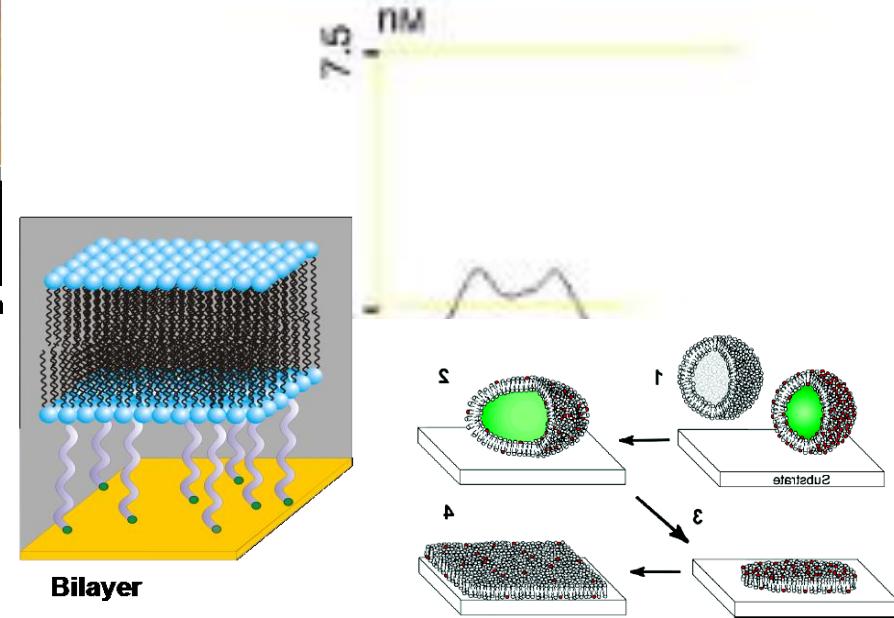
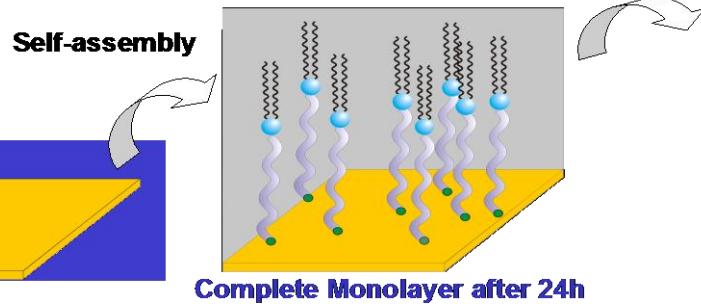
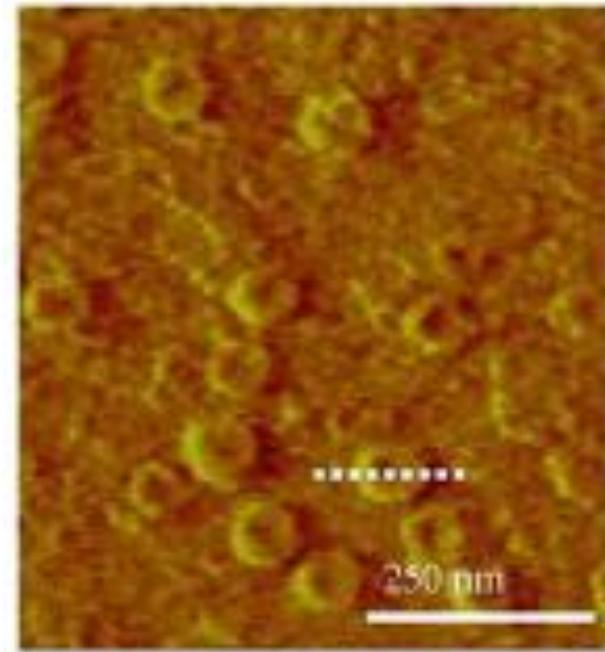
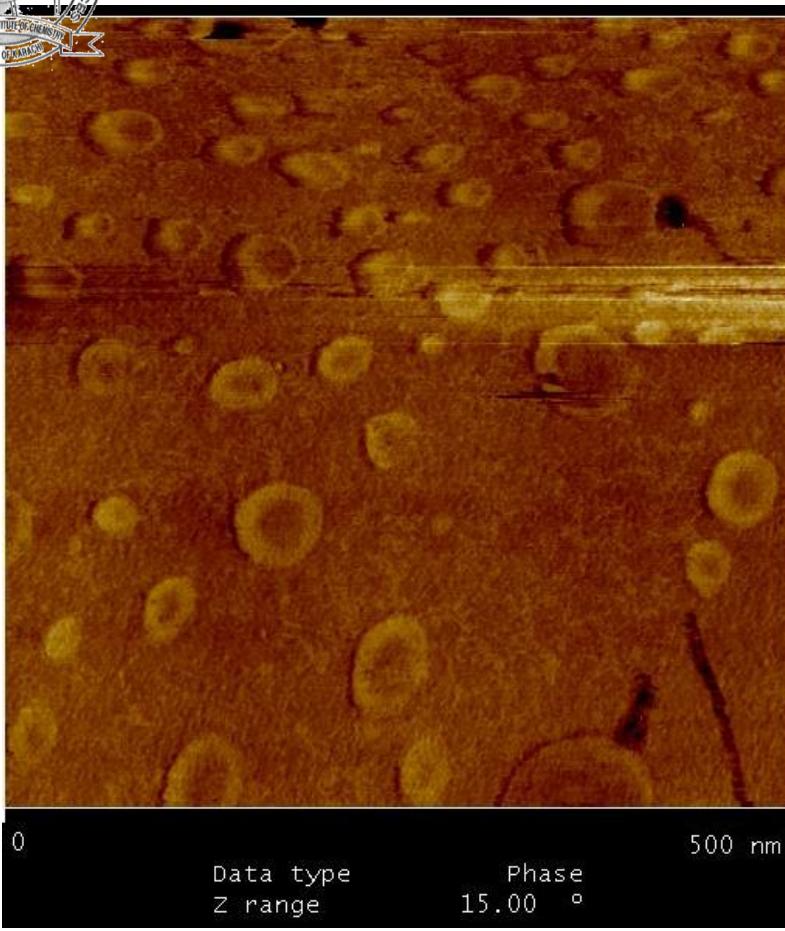
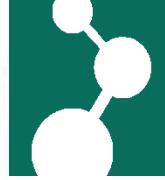
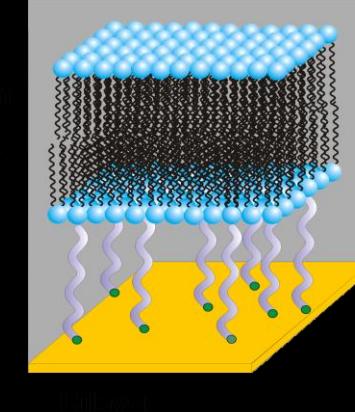
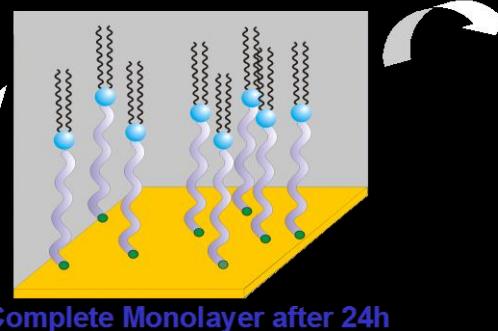


Figure 1. Polydispersed DPhyPC vesicles self-assemble on a gold substrate. After 10 minutes air dried, vesicles form a bilayer (1) which undergoes fusion to form a complete monolayer (2). After 24 hours the bilayer disassembles to form vesicles (3) which undergo further fusion to form larger vesicles (4).

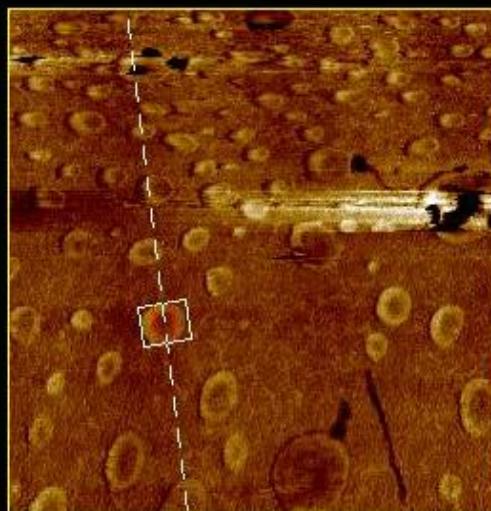
3 µg/ml solutions of DPhyPC vesicles after 20 minutes.



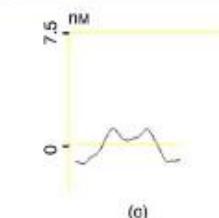
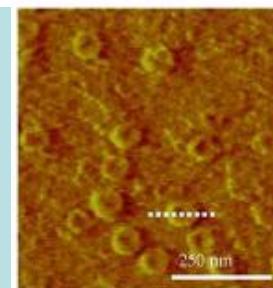
Section Analysis



L	56.02 nm
RMS	0.107 °
1c	DC
Ra(1c)	0.069 °
Rmax	0.363 °
Rz	0.253 °
Rz Cnt	6
Radius	24.065
Sigma	10.808

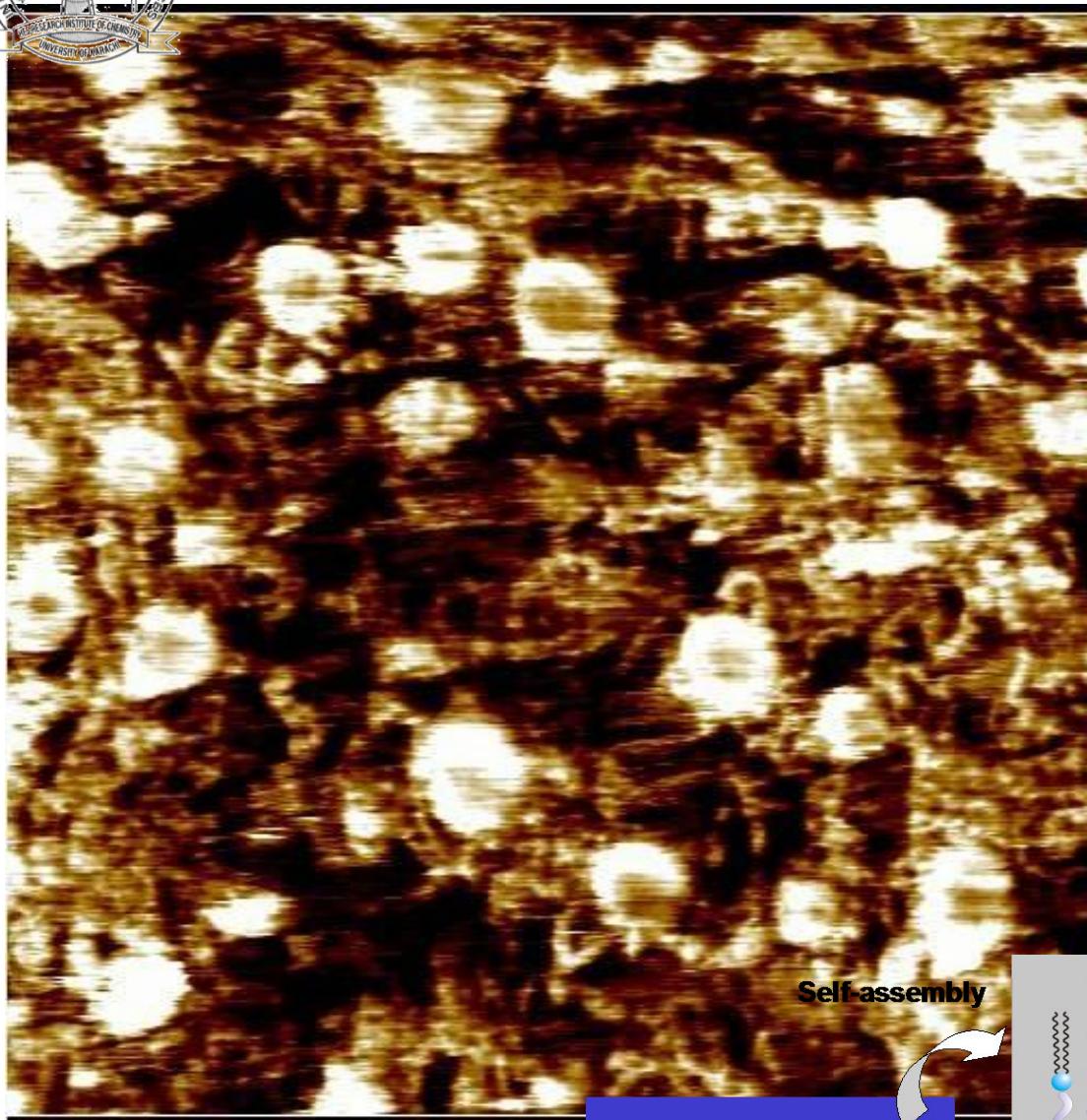


Colloids and Surfaces B: Biointerfaces 34 (2004) 41-51



Surface distance	72.26
Horiz distance(L)	66.02 nm
Vert distance	0.126 °
Angle	
Surface distance	
Horiz distance	
Vert distance	
Angle	
Surface distance	
Horiz distance	
Vert distance	
Angle	
Spectral period	DC
Spectral Freq	0 /µm

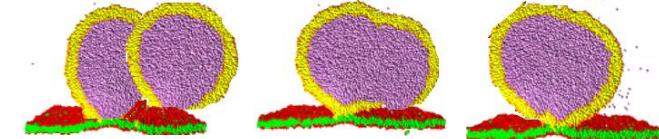
3 µg/ml solutions of DPhyPC vesicles after 1hour.



Au Substrate

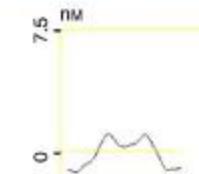
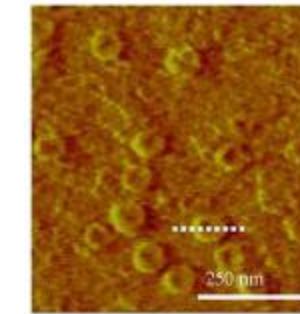
Complete Monolayer after 24h

Formation of Giant Vesicle

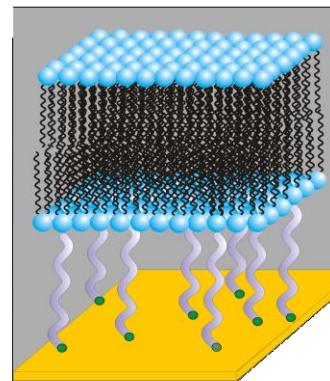
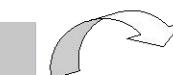


Colloids and Surfaces B: Biointerfaces 34 (2004) 41–51

(c) of EggPC vesicles. The images were captured in pure water and with initial

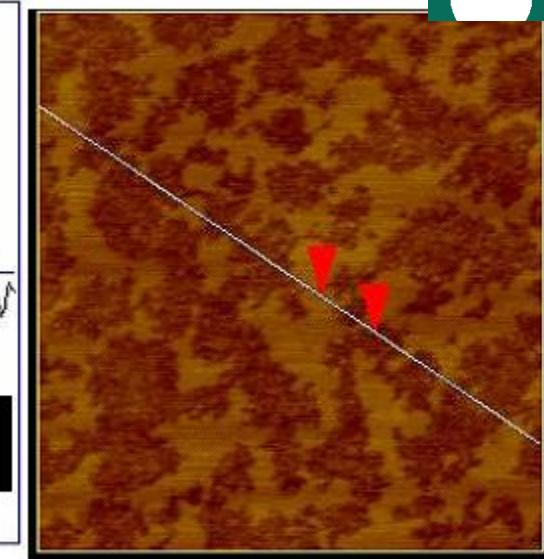
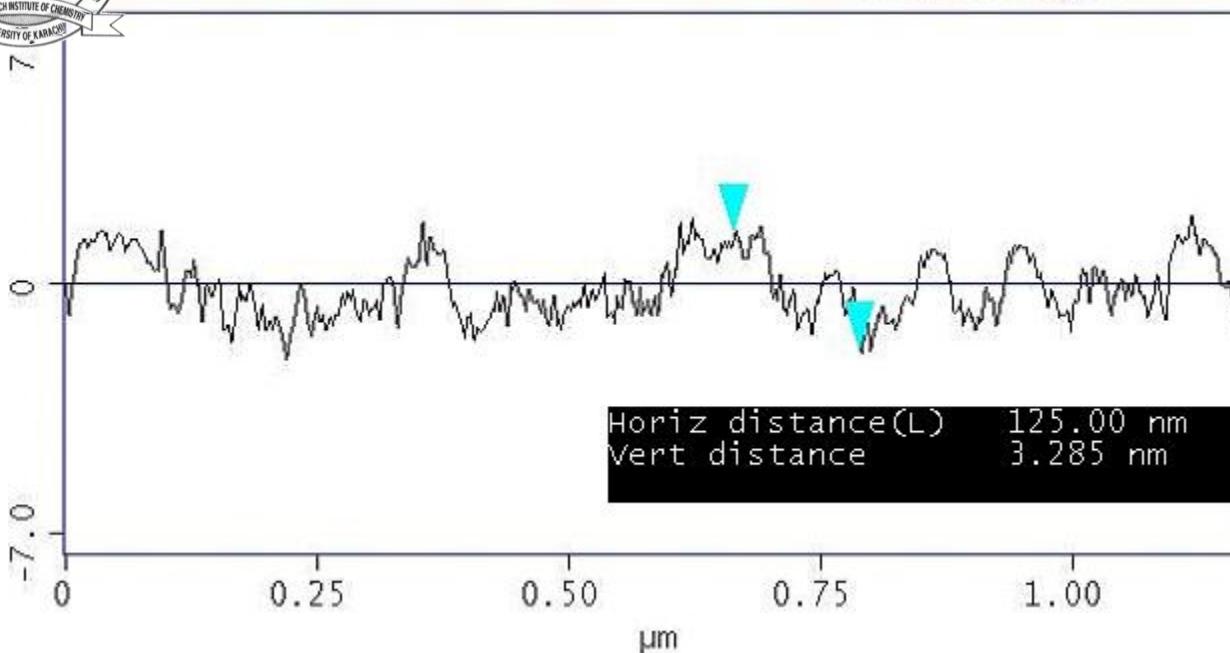


(c) Vesicle fusion

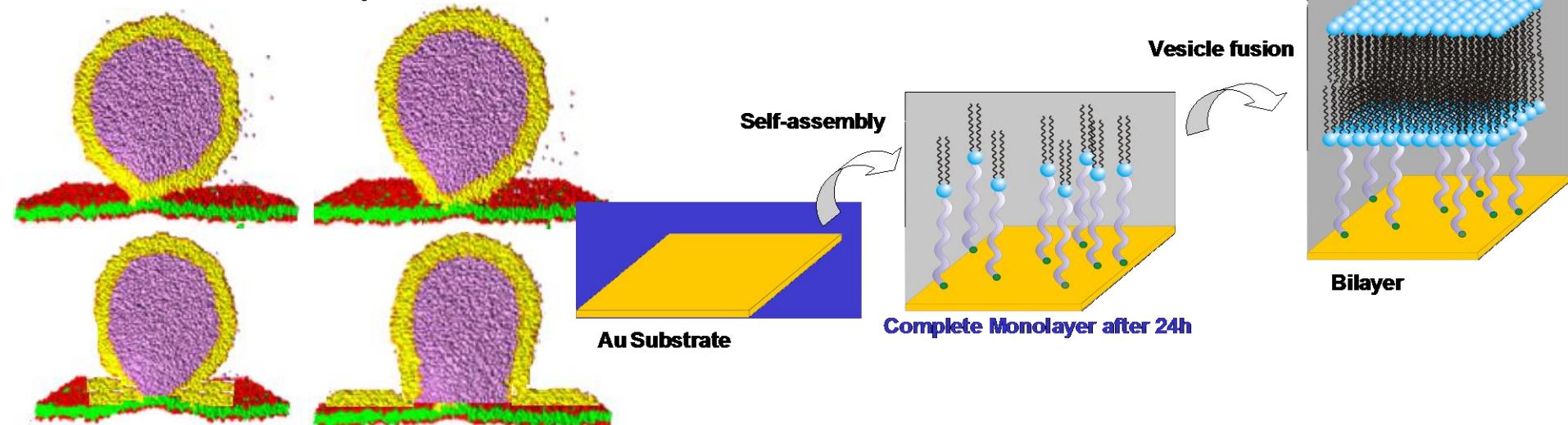


3 µg/ml solutions of DPhyPC vesicles after 2.5hours.

Section Analysis

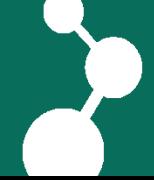


Mechanism of Bilayer formation

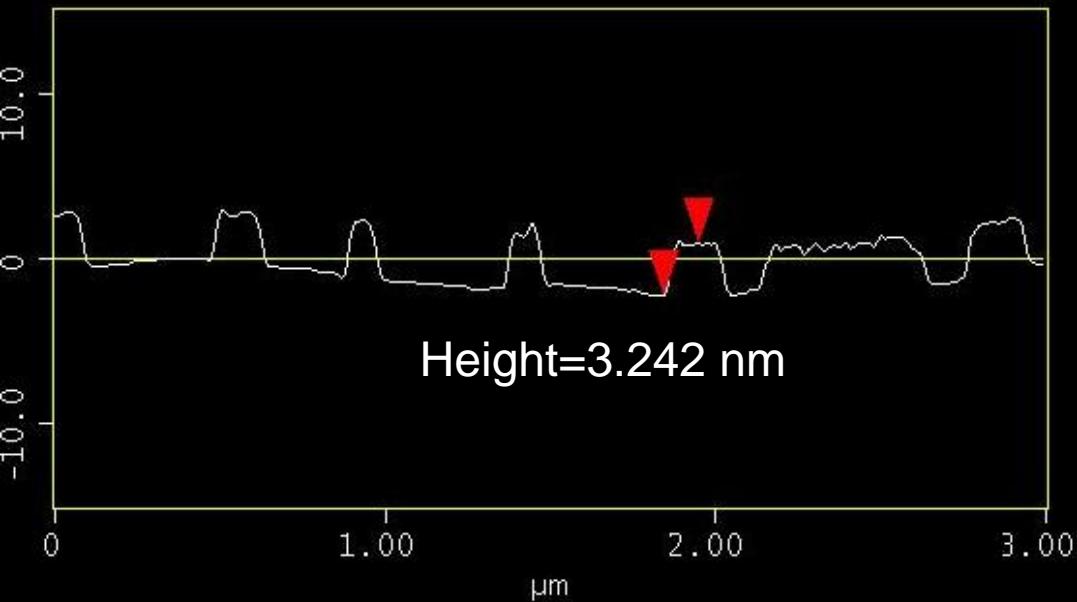




3 µg/ml solutions of DPhyPC vesicles after 3 hours.



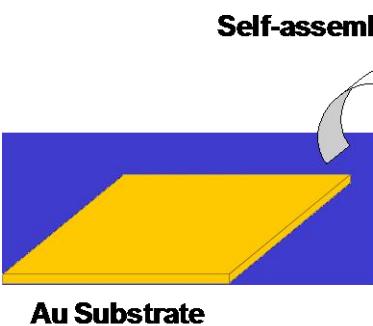
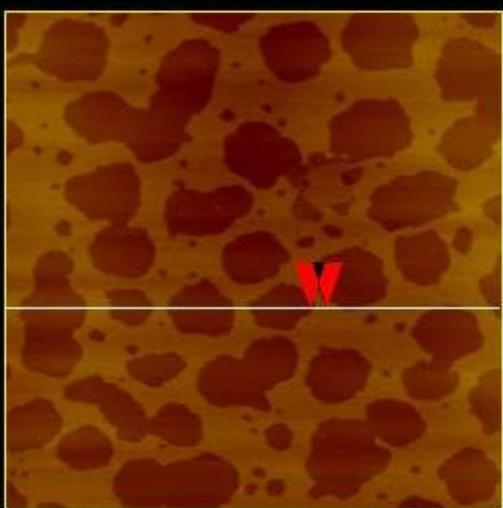
Section Analysis



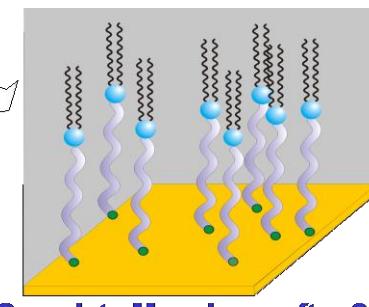
L	105.47 nm
RMS	1.075 nm
1c	DC
Ra(1c)	0.566 nm
Rmax	2.280 nm
Rz	2.280 nm
Rz Cnt	2
Radius	577.69 nm
Sigma	0.480 nm

Spectrum

Surface distance	105.66 nm
Horiz distance(L)	105.47 nm
Vert distance	3.242 nm

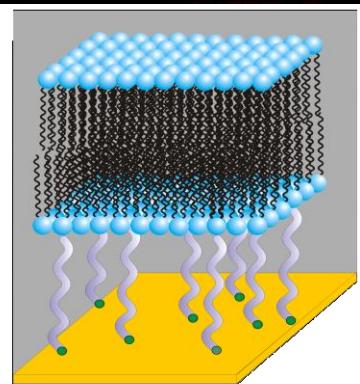


Self-assembly



Complete Monolayer after 24h

Vesicle fusion



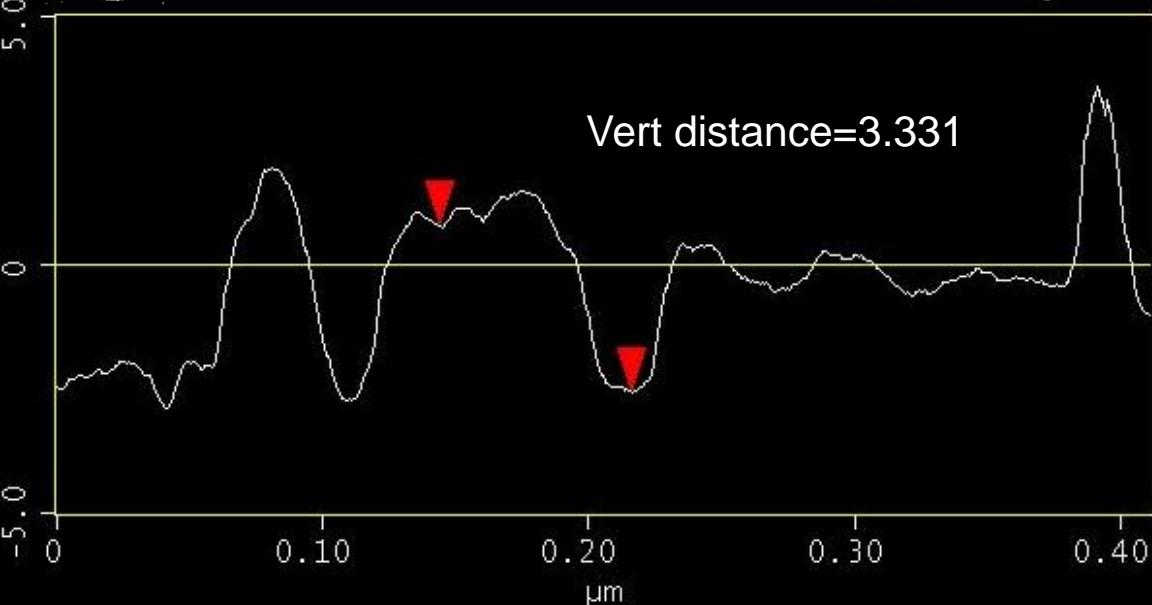
Bilayer



3 µg/ml solutions of DPhyPC vesicles after 6hours.



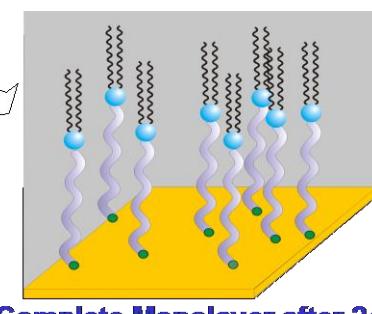
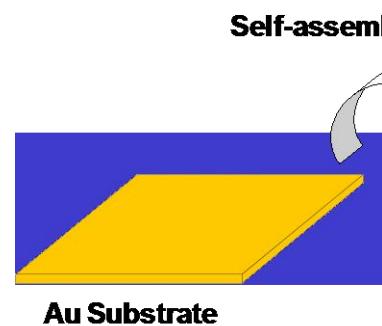
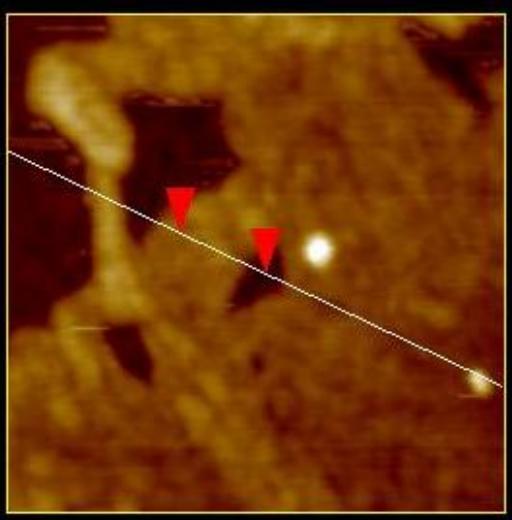
Section Analysis



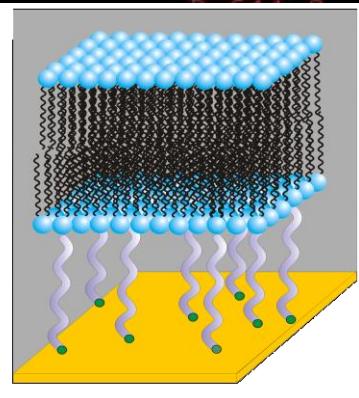
L	72.132 nm
RMS	1.425 nm
Tc	DC
Ra(Tc)	0.754 nm
Rmax	2.615 nm
Rz	2.615 nm
Rz Cnt	2
Radius	221.35 nm
Sigma	0.363 nm

Spectrum

Surface distance 72.601 nm
Horiz distance(L) 72.132 nm
Vert distance 3.331 nm



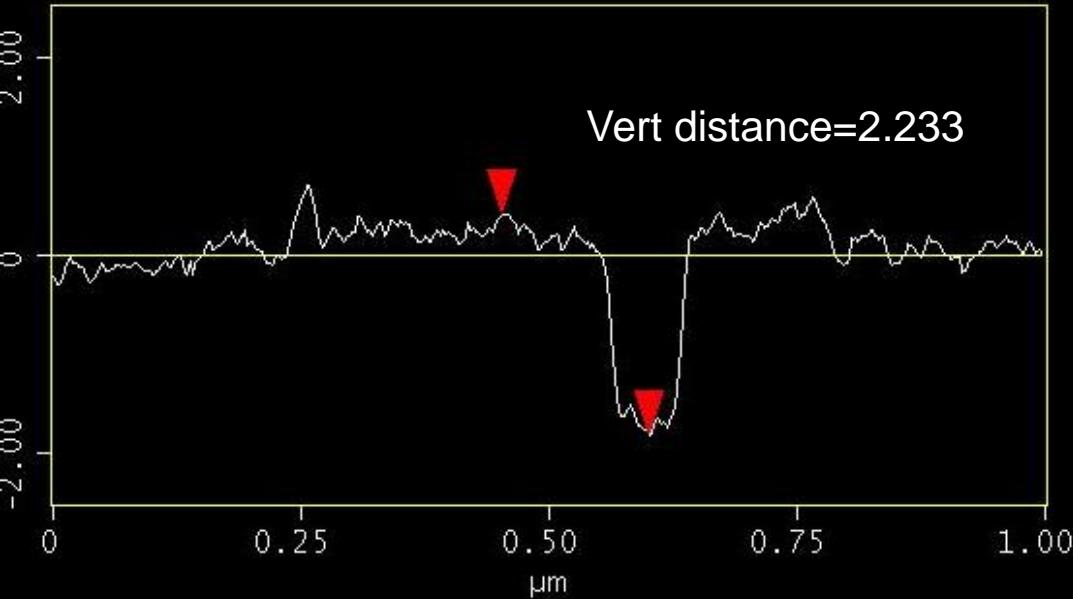
Vesicle fusion



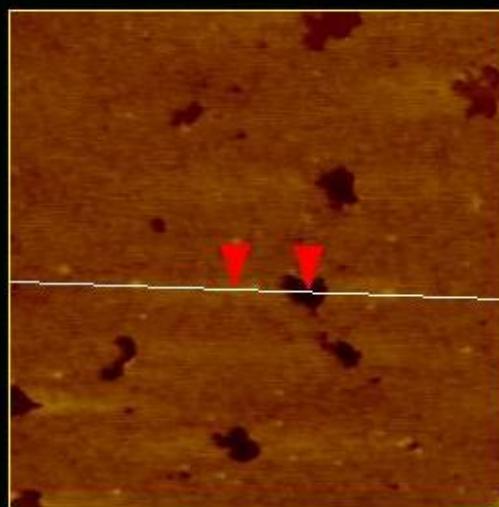


3 µg/ml solutions of DPhyPC vesicles after 8hours.

Section Analysis

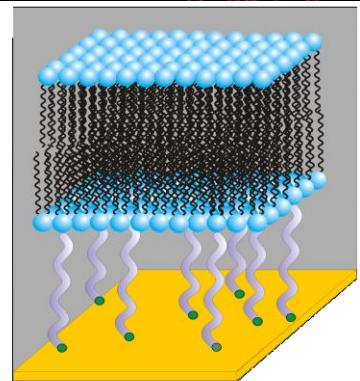
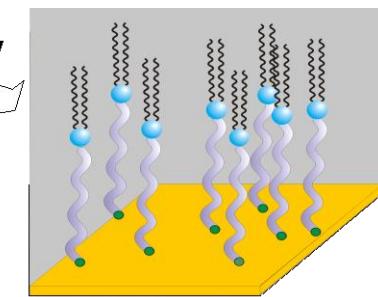
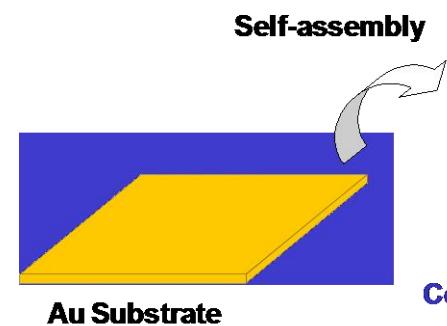


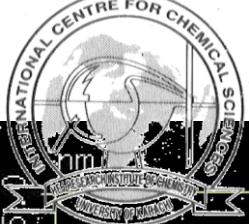
Surface distance 148.55 nm
 Horiz. distance(L) 148.44 nm
 Vert. distance 2.233 nm
 Rmax 1.315 nm
 Rz 1.315 nm
 Rz Cnt 2
 Radius 1.597 µm
 Sigma 0.330 nm



Spectrum

Vesicle fusion



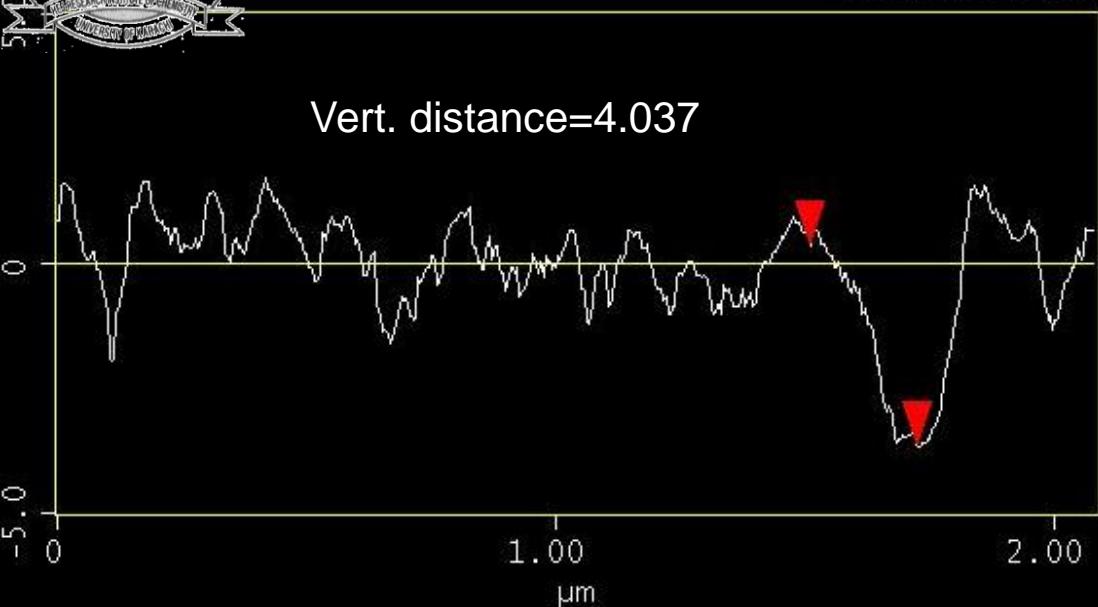


3 µg/ml solutions of DPhyPC vesicles after 9hours.



Section Analysis

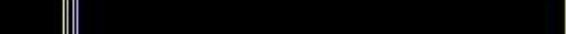
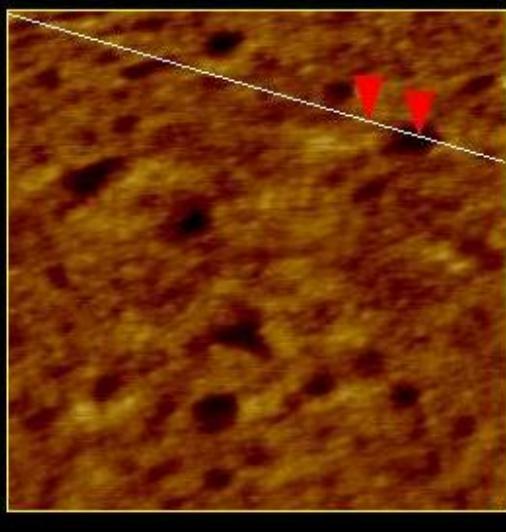
Vert. distance=4.037



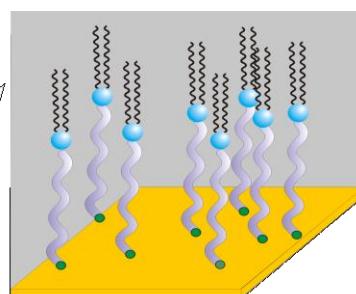
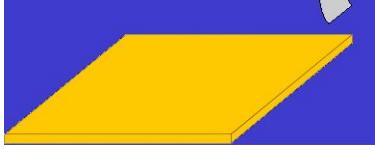
L	214.84 nm
RMS	1.475 nm
1c	DC
Ra(1c)	0.277 nm
Rmax	1.363 nm
Rz	0.577 nm
Rz Cnt	valid
Radius	1.938 µm
Sigma	0.860 nm

Spectrum

Surface distance	215.08 nm
Horiz. distance(L)	214.84 nm
Vert. distance	4.037 nm
Angle	1.077 °

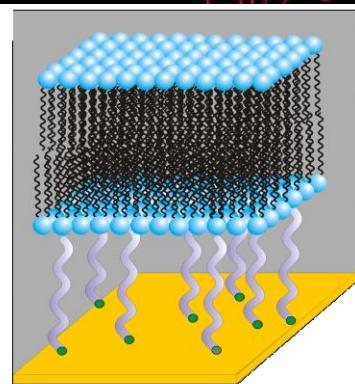


Self-assembly



Complete Monolayer after 24h

Vesicle fusion



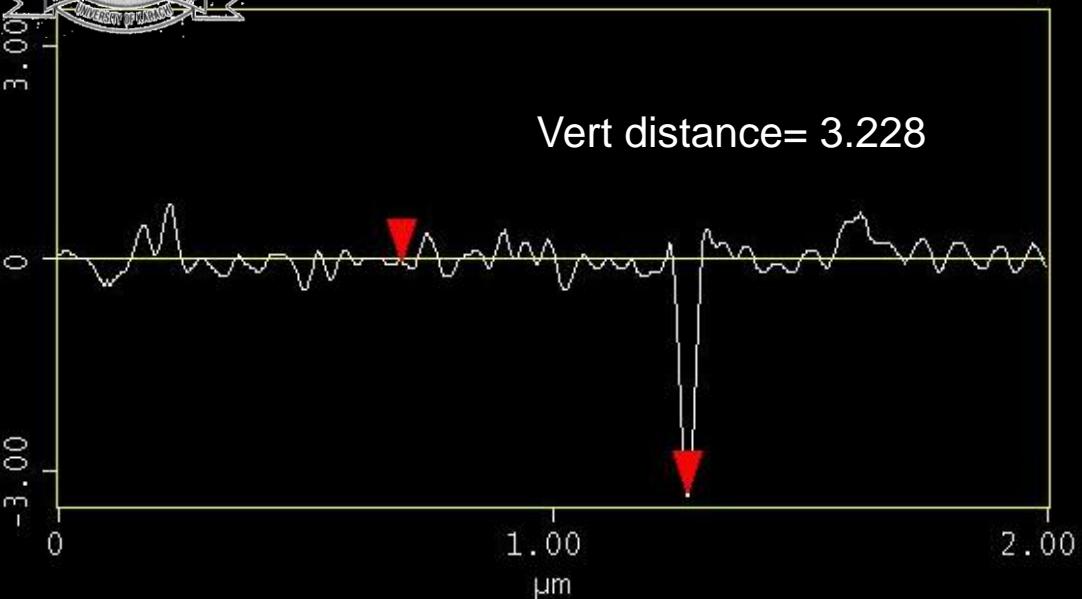
Bilayer



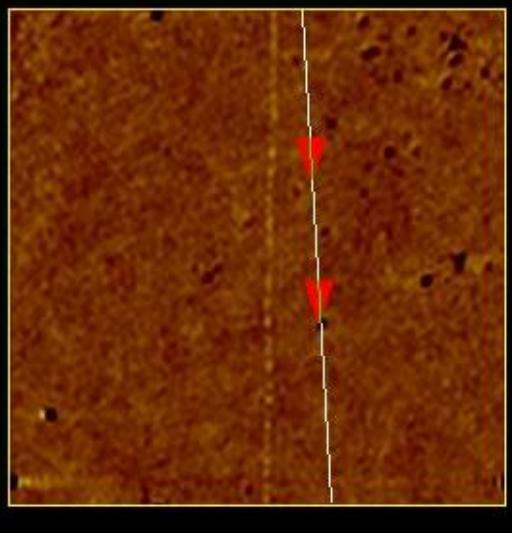
3 µg/ml solutions of DPhyPC vesicles after 12hours.



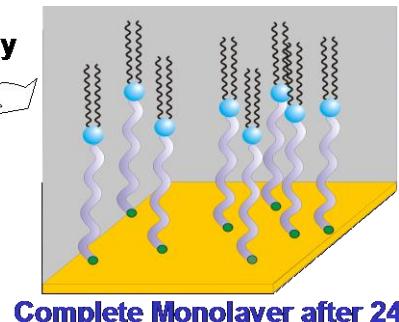
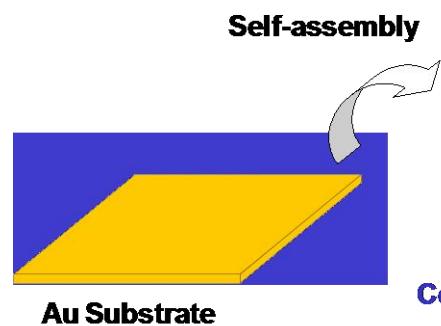
Section Analysis



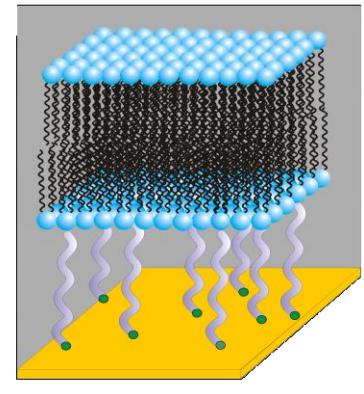
Surface distance 578.41 nm
Horiz distance(L) 578.13 nm
Vert distance 3.288 nm
Avg Z 0.325 nm



Spectrum



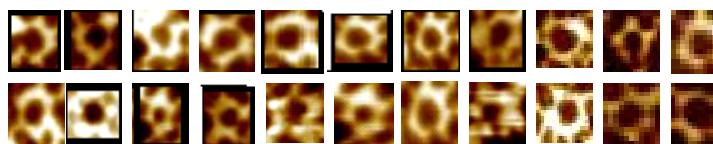
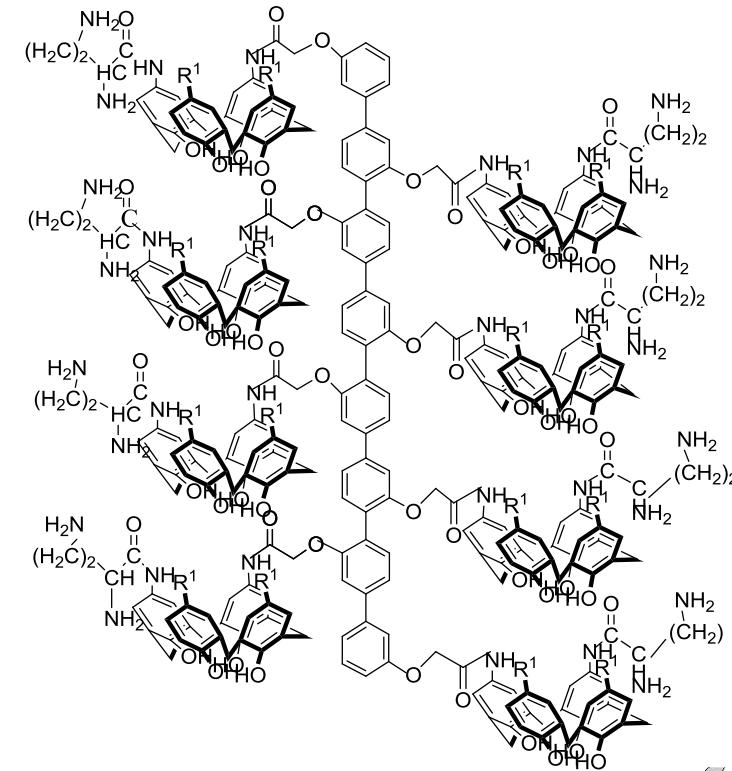
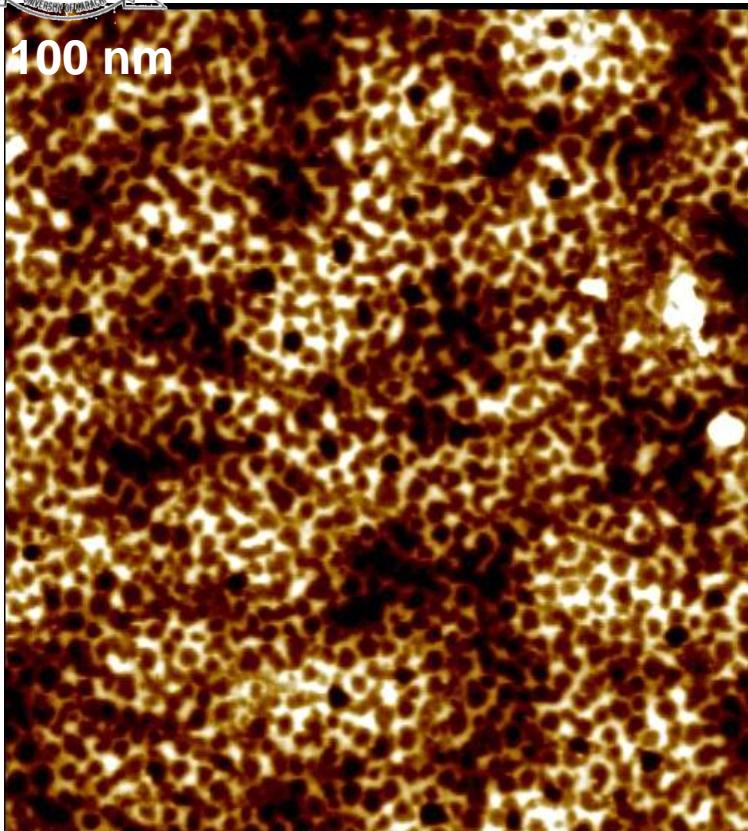
Vesicle fusion



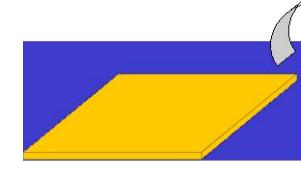
Bilayer

0.005µg/µl solution of Calixarene pore after 08 hours

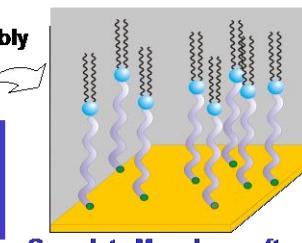
Taking 02 µl solution from a stock of 0.5µg/µl and adding it to 400 µL



Self-assembly

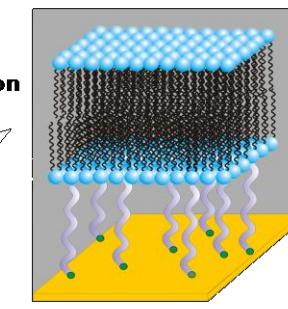


Au Substrate

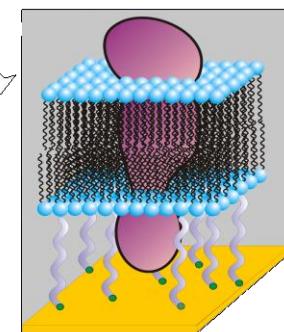


Complete Monolayer after 24h

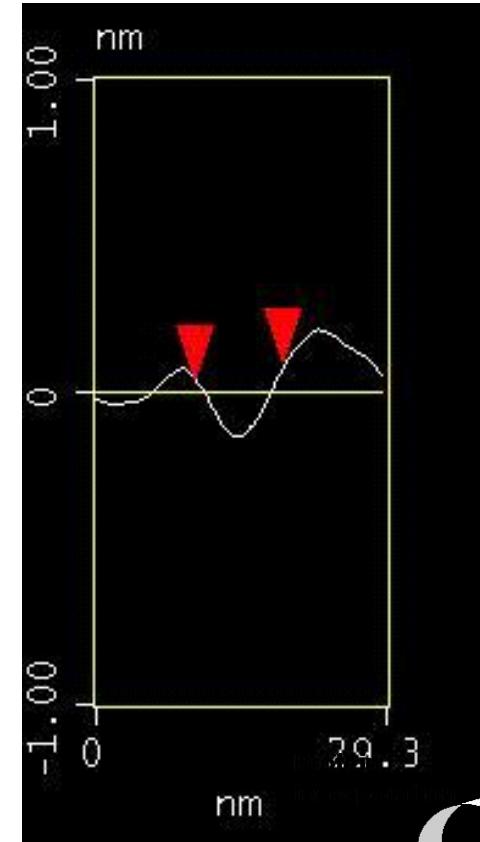
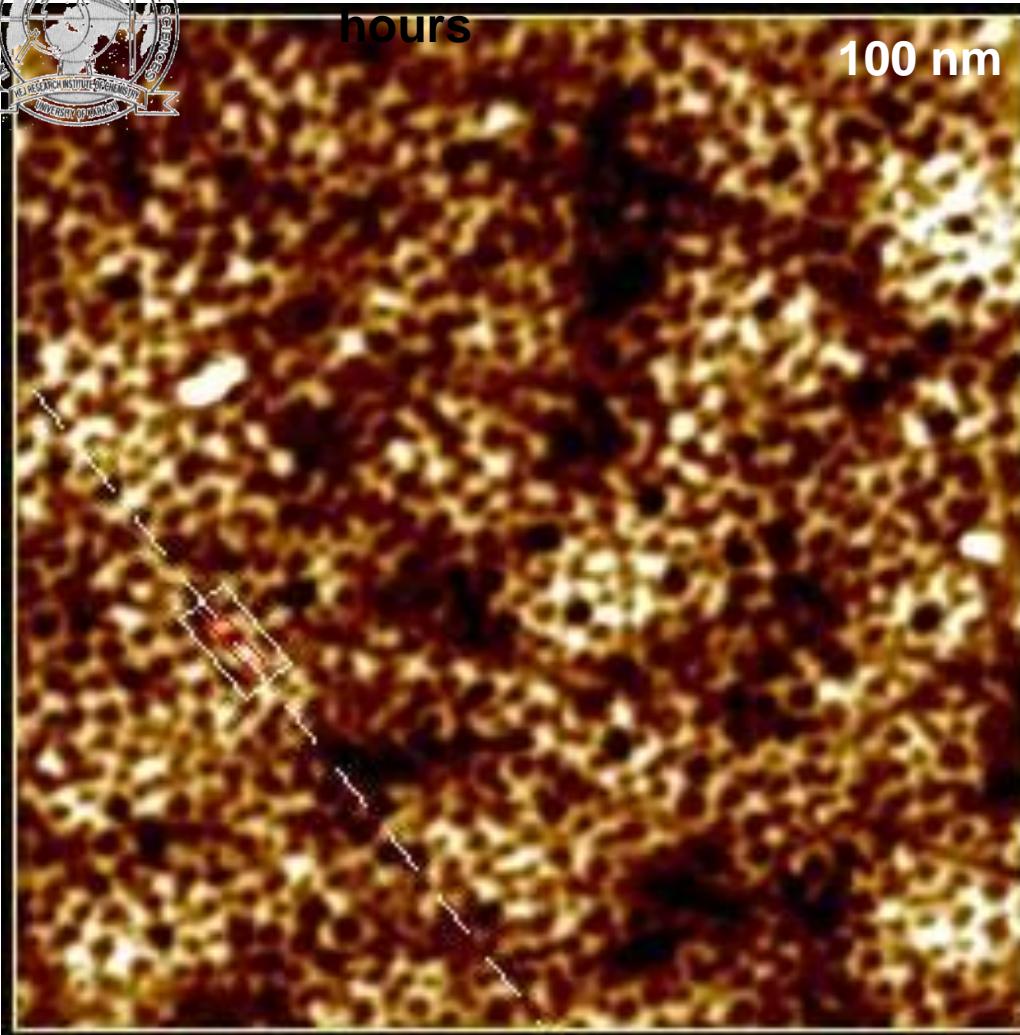
Vesicle fusion



Bilayer

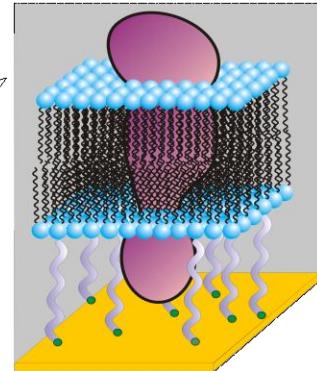
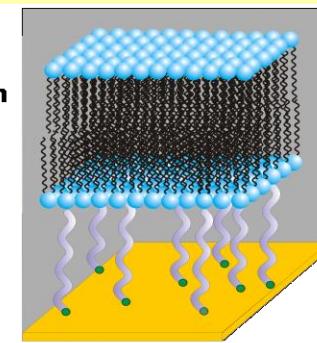


0.005 μ g/ μ l solution of Calixarene pore after 08 hours

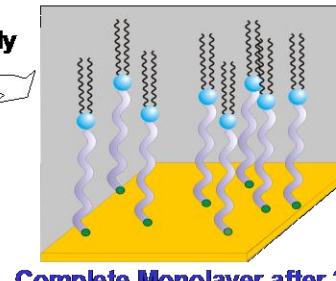


External diameter=8.858 nm

Vesicle fusion



Self-assembly



Complete Monolayer after 24h

Bilayer

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